



Intermediate Temperature Direct Ammonia Fuel Cells

Jason C. Ganley

Howard University
Department of Chemical Engineering
Washington, DC



Ammonia for Fuel Cells



CH4 103 (1.5 H2)



- Very mild enthalpy of reforming
- NH_3 is a liquid at room temperature and 10 bar
 - Power density is comparable to other liquid fuels
 - Vaporizes when throttled (no flash line required)
- Essentially non-flammable, non-explosive
- 171 kWh of motive power from 57 litres ammonia (38 kg) with 48% efficient fuel cell system incl. motor
- Highway driving: 19 kW; yields 9 hours of cruising
- 65 miles per hour takes you 585 miles (~940 km)

356 kWh nom
150 mi gas

Fuel Cells or ICE?

• Internal combustion engine

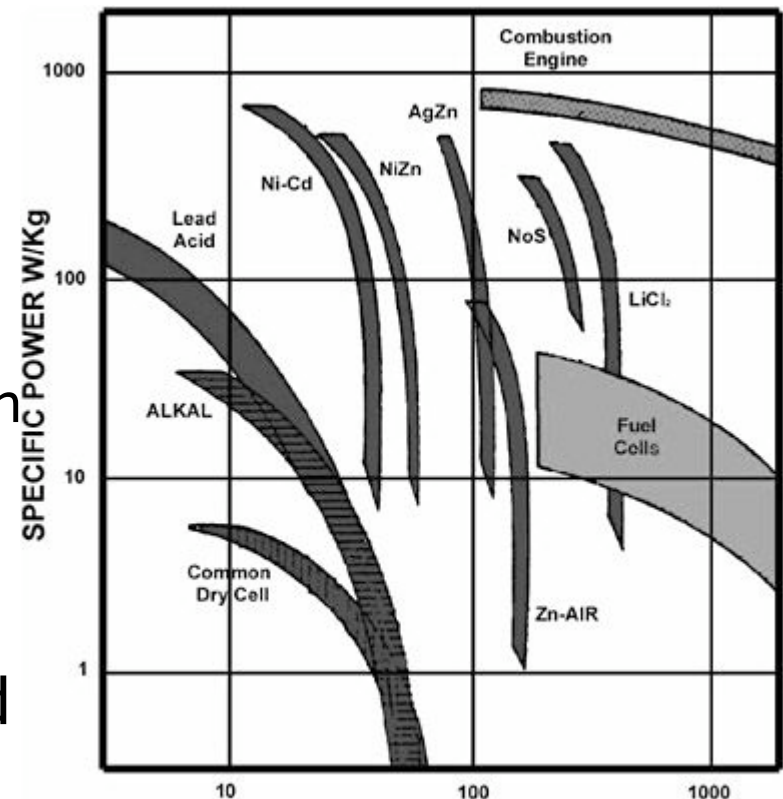
- Rated at peak hp
- My old car: 85 hp Geo Metro
 - ◆ Cruising uses only about 10 hp
 - ◆ I, however, drive like an old lady
- Peak hp is only for short periods of time, even in military apps
- Maximum torque only at certain rpm

• The electric motor

- Rated for continuous power
- Can provide **at least** 5X rated power intermittently
- Maximum torque available from a dead stop (0 rpm)



← Electrochemical Capacitors



SPECIFIC ENERGY WH/Kg
 Board on Army Science
 And Technology (BAST)



Fuel Comparison



Fuel	Base MJ/liter	Reformed MJ/liter*
H ₂ (5000 psia)	4.0	4.0
H ₂ (liq.)	9.9	9.9
NH ₃ (liq.)	15.3	13.6
Methanol	17.9	10.2
Ethanol	23.4	9.1
Propane (liq.)	29.4	8.6
Gasoline	36.2	9.2
JP-8	40.5	9.7

* Includes heat and water volume required for steam reforming



Operating Temperature: A Key Characteristic



- **Low Temperature Fuel Cell Advantages**

- Quick start-up to operating temperature ($\sim 100^{\circ}\text{C}$)
- Wide range of cell construction materials

- **High Temperature Fuel Cell Advantages**

- Fuel flexibility via internal fuel reforming
- Inexpensive, base metal electrocatalysts
- Easier heat recovery for increased efficiency

- **Intermediate Temperature Fuel Cells: The Best of Both Worlds?**

- Precious metal catalysts not needed above $\sim 300^{\circ}\text{C}$
- Steel internals may be used below $\sim 500^{\circ}\text{C}$



Contemporary Fuel Cell Options



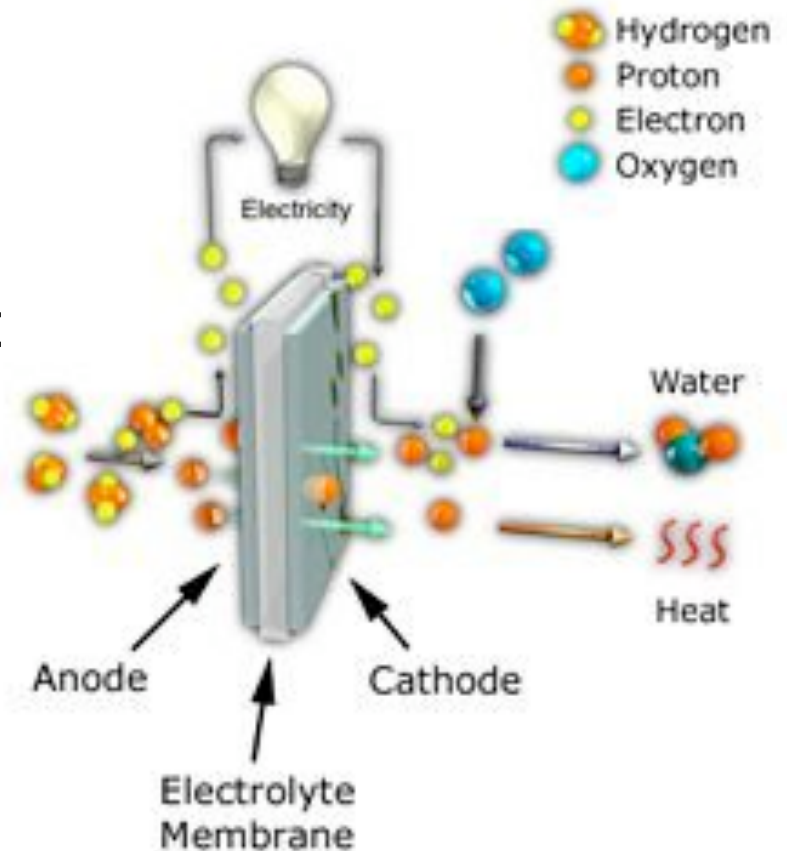
- **Polymer Electrolyte Membrane Fuel Cells (PEMFC) [80°C, H⁺]**
- **Alkaline Fuel Cells (AFC) [150°C, OH⁻]**
- **Phosphoric Acid Fuel Cells (PAFC) [220°C, H⁺]**
- **Intermediate Temp Fuel Cell (ITFC) [500°C, H⁺]**
- **Protonic Ceramic Fuel Cells (PCFC) [550°C, H⁺]**
- **Molten Carbonate Fuel Cells (MCFC) [650°C, CO₃²⁻]**
- **Solid Oxide Fuel Cells (SOFC) [900°C, O²⁻]**
- **Solid Oxide Fuel Cells (SOFC) [900°C, O²⁻]**

- **External NH₃ reforming required**

- Trace NH₃ poisons acid membrane
- Even well-reformed fuel must be scrubbed

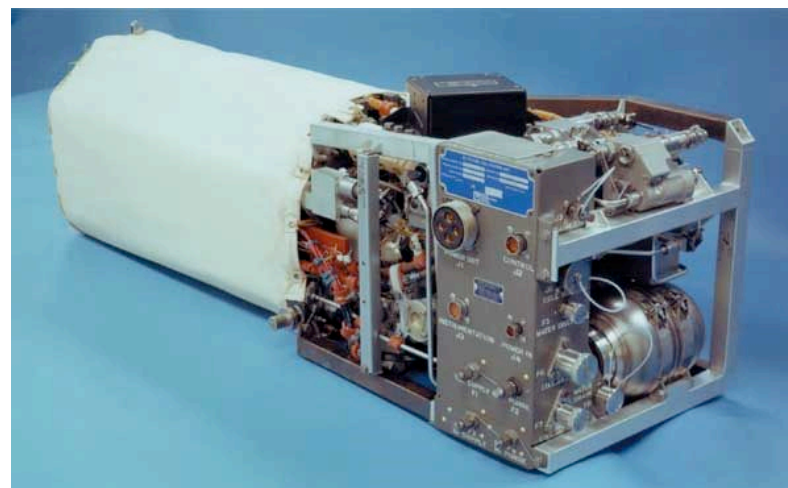
- **Temperature and humidity issues**

- Above 100°C, membrane dries out, loses conductivity
- Inefficient heat recovery
- Precious metal electrocatalysts required



Plug Power Inc.

- **AFCs are tolerant of residual NH_3**
 - Still must crack NH_3 externally
 - Precious metal catalysts
- **Lifetime issues**
 - Corrosive electrolyte
 - Intolerant of CO_2
 - ◆ Formation of carbonate precipitates
 - ◆ Fuel and air streams must be purified
 - Expensive immobilization matrix for electrolyte



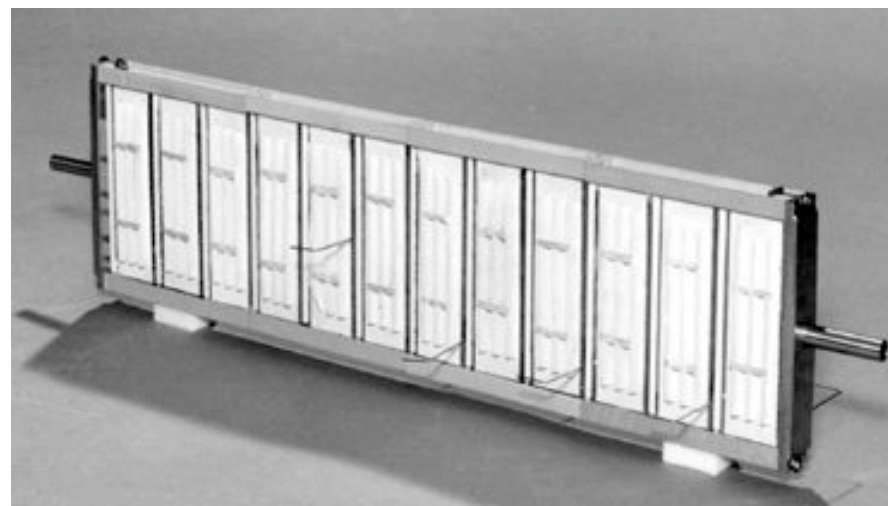
United Technologies

- **MCFC/ NH_3 is a fair match**

- Direct-ammonia capable
- Efficient heat recovery
- Cheaper electrocatalysts
- Liquid electrolyte, good σ

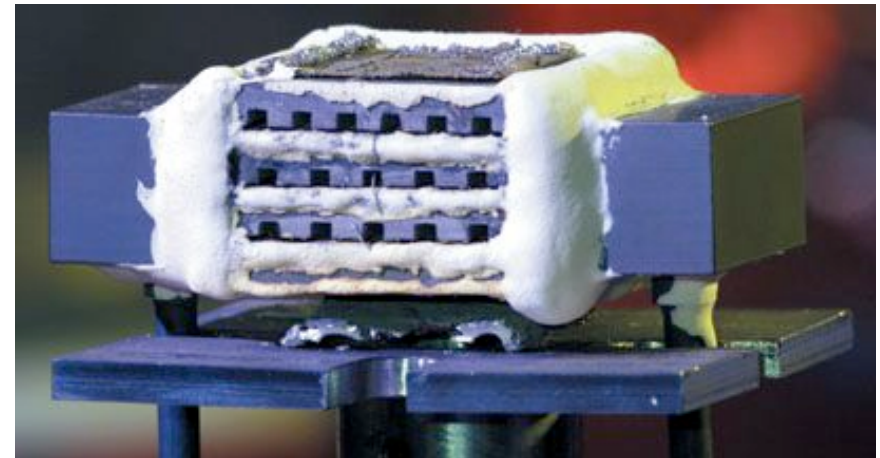
- **Problems with MCFCs**

- Electrolyte is very corrosive to FC components
- Carbon dioxide recycling
- Not suited to intermittent use
- Ammonia crossover reduces system efficiency significantly



United States Army

- **SOFC/NH₃ is a very good match**
 - Direct-ammonia capable
 - Efficient heat recovery
 - Ni/cermet electrocatalysts
 - Solid, durable electrolyte
- **Problems with SOFCs**
 - Extreme operating temperature not conducive to mobile applications
 - NO_x formation at anode
 - Fuel is diluted by product water



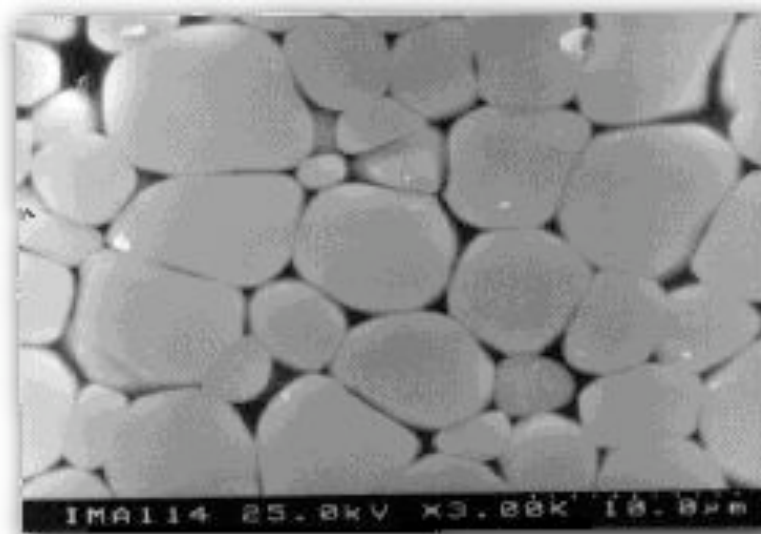
Lawrence Livermore National Lab

- **PCFC/NH₃ is an excellent match**

- Direct-ammonia capable
- Efficient heat recovery
- Ni/cermet electrocatalysts
- Solid, durable electrolyte

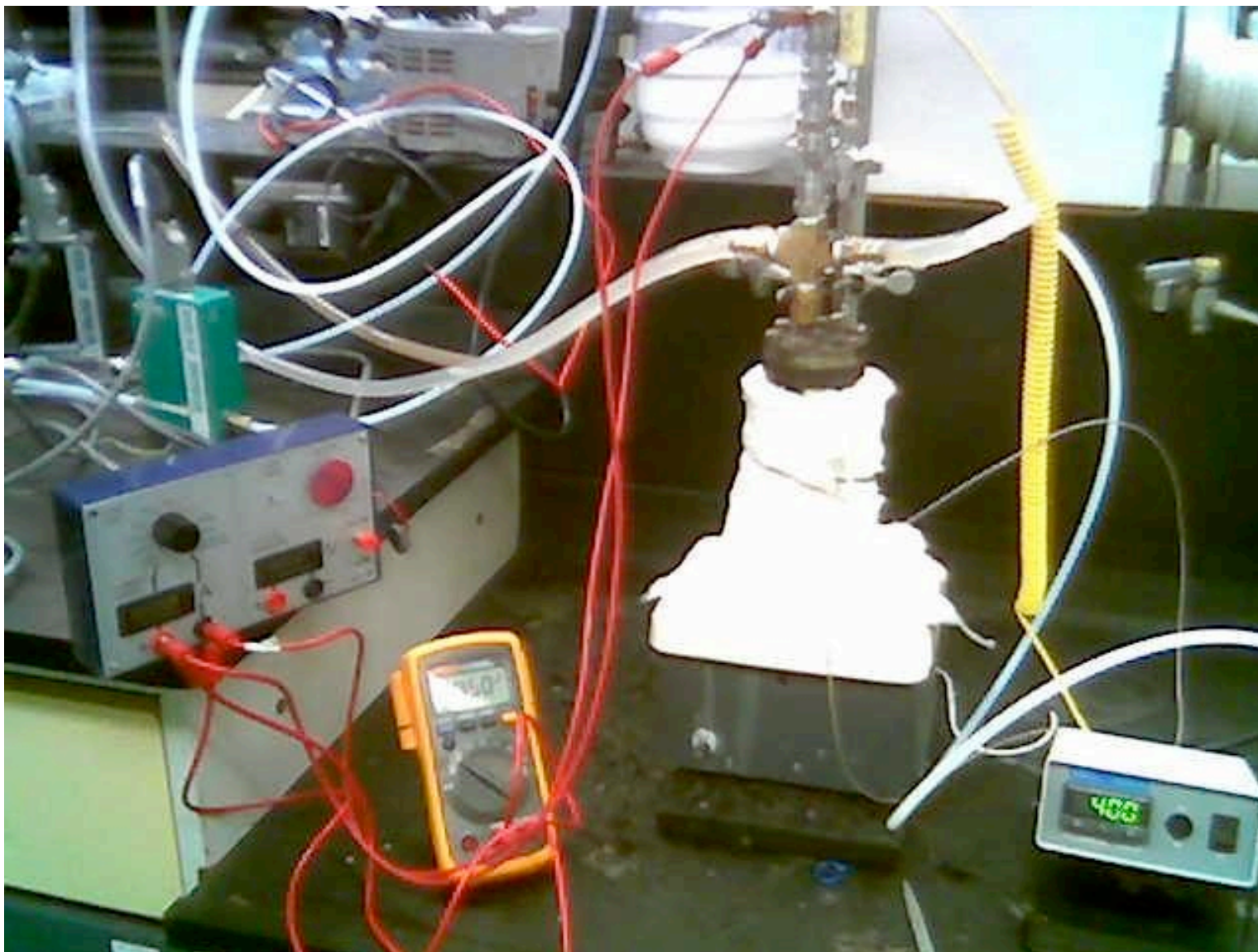
- **Overcomes SOFC limitations**

- Lower operating temperature allows stainless steel internals
- No NO_x formation at anode
- Fuel not diluted by product water
- Complete ammonia conversion possible



University of Aviero, Portugal

Filling the Temperature Gap: Intermediate Temperature DAFC

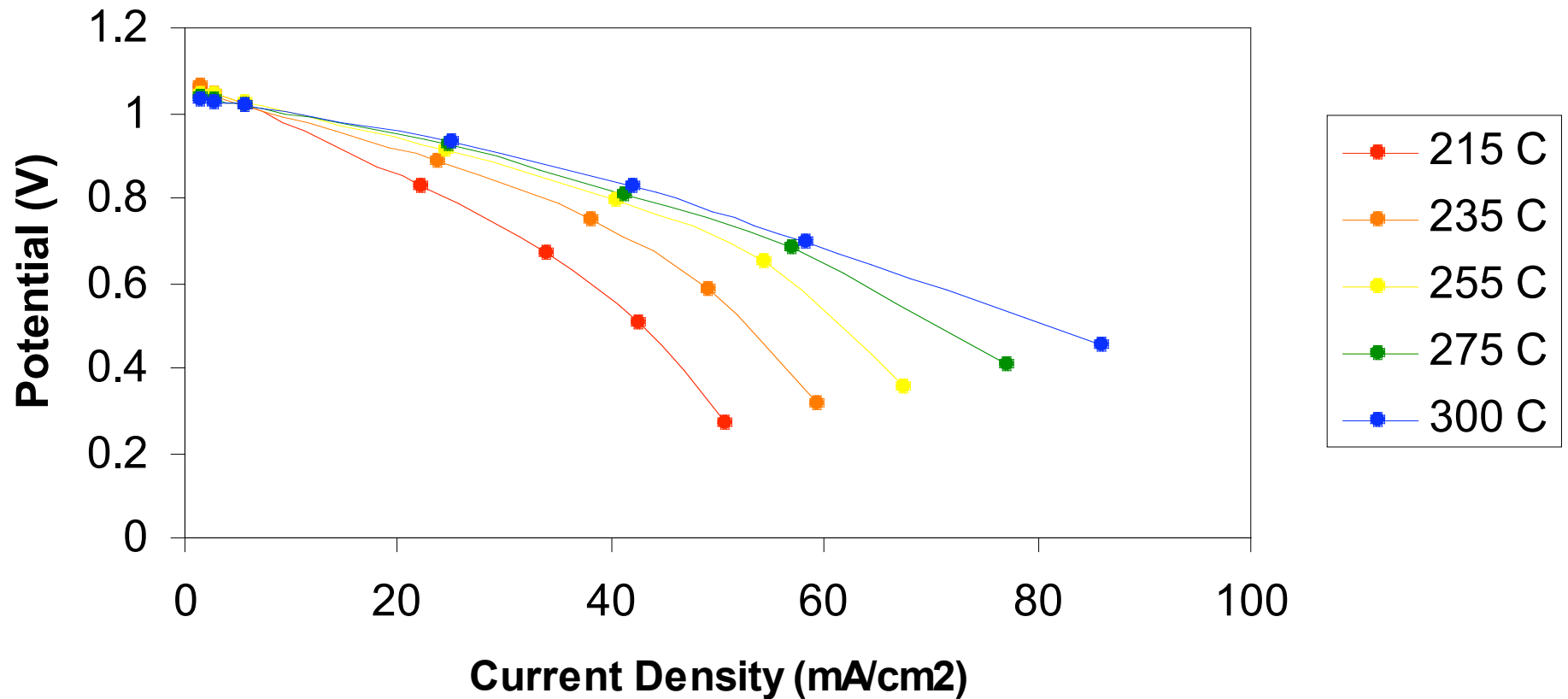




Hydrogen Cell Performance



Hydrogen Fuel

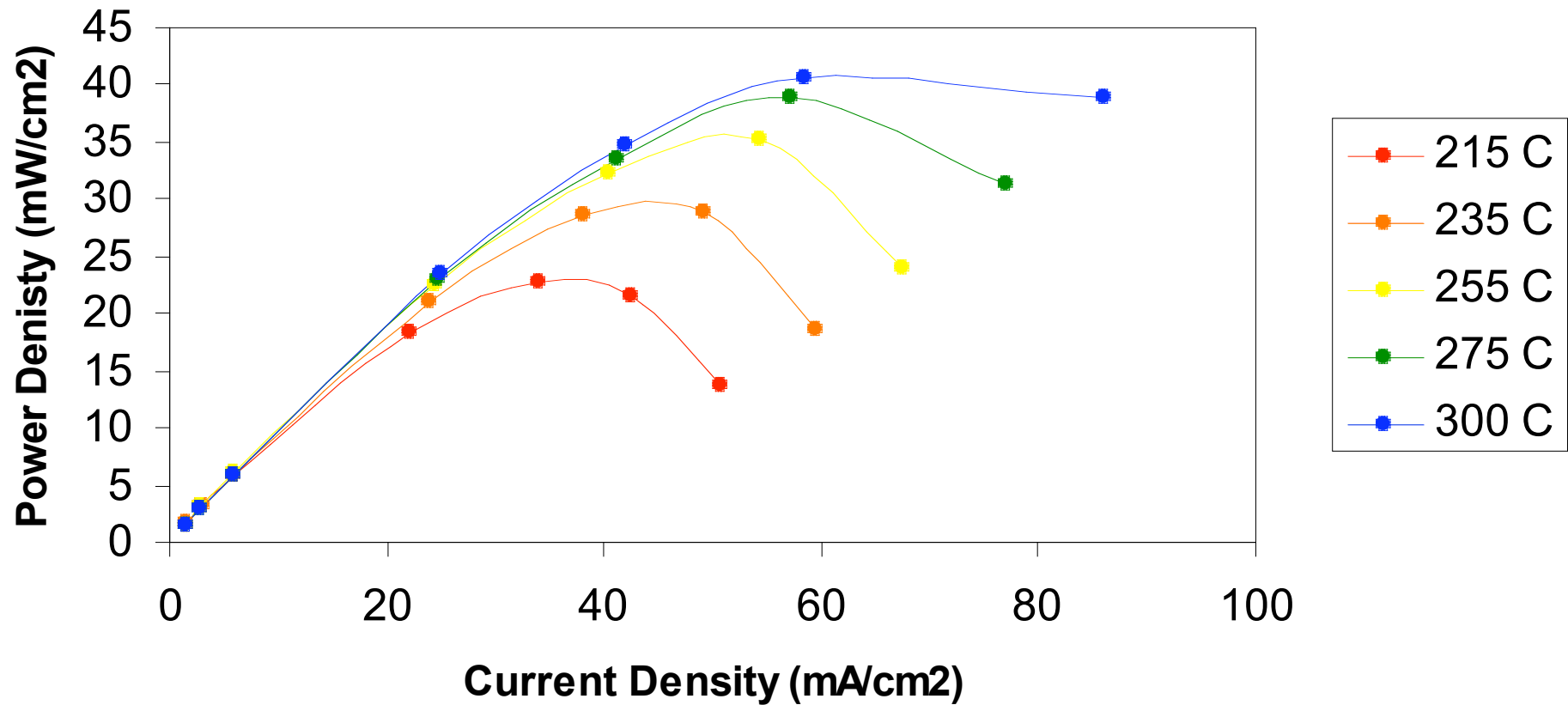




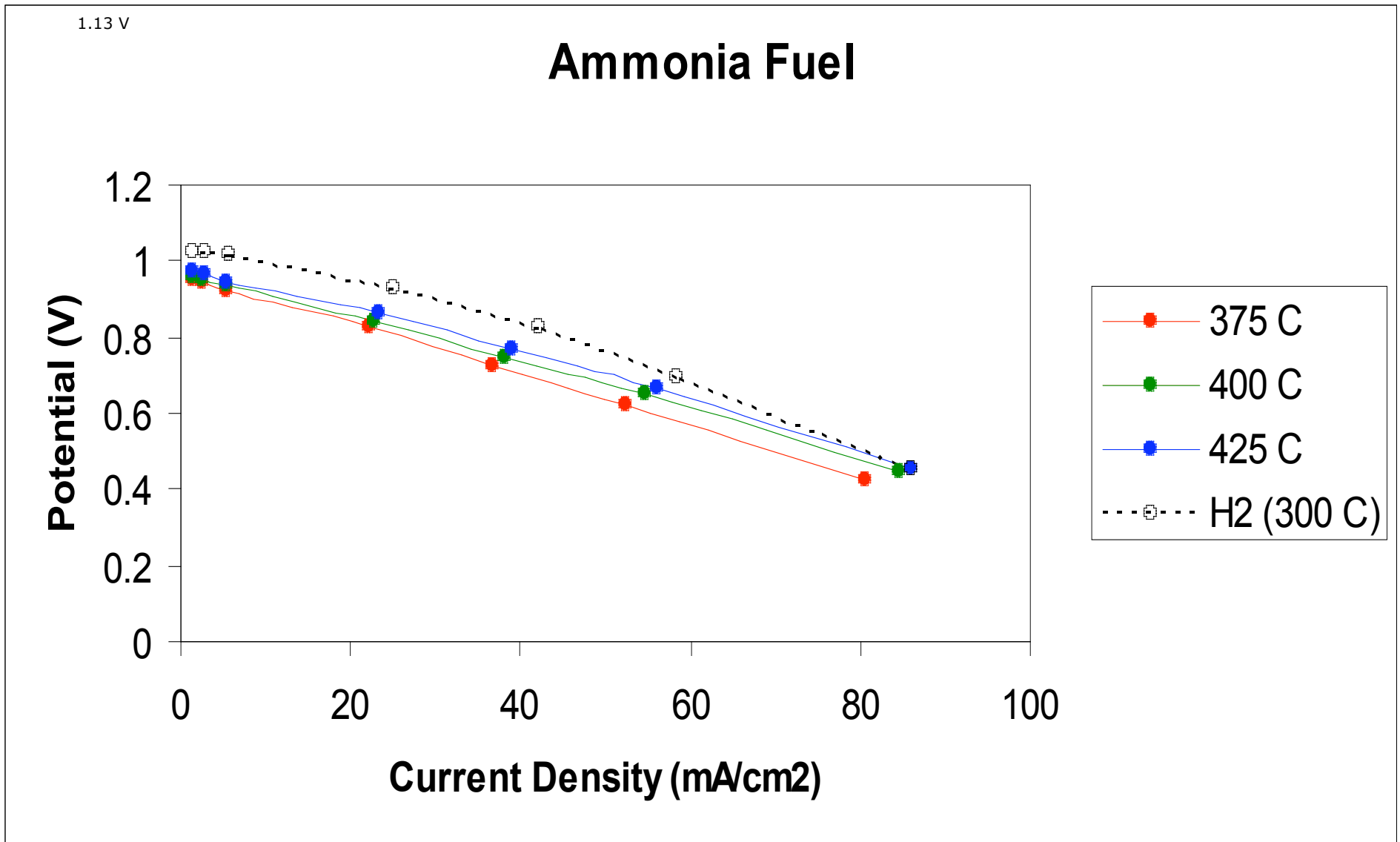
Hydrogen Cell Performance



Hydrogen Fuel

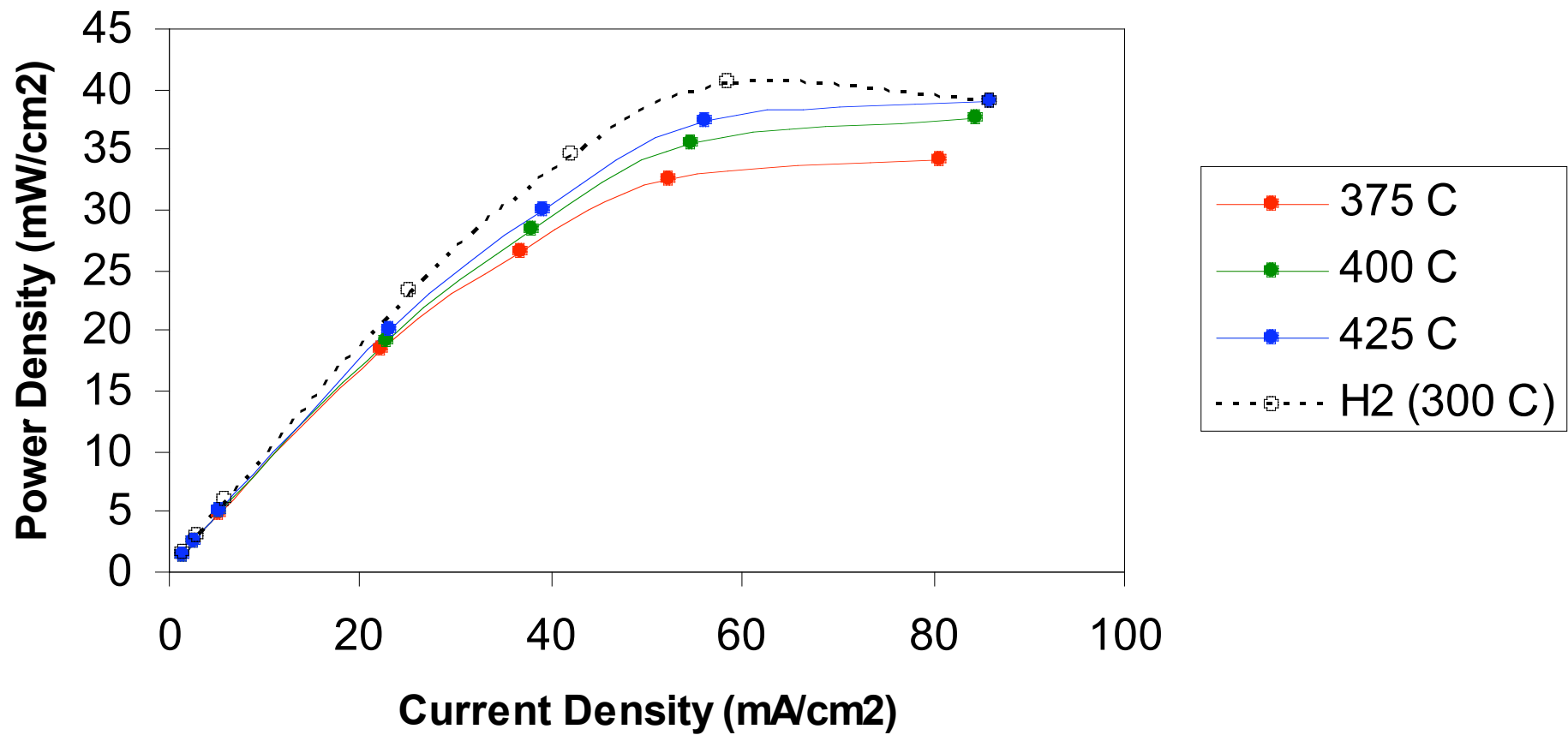


Ammonia Cell Performance

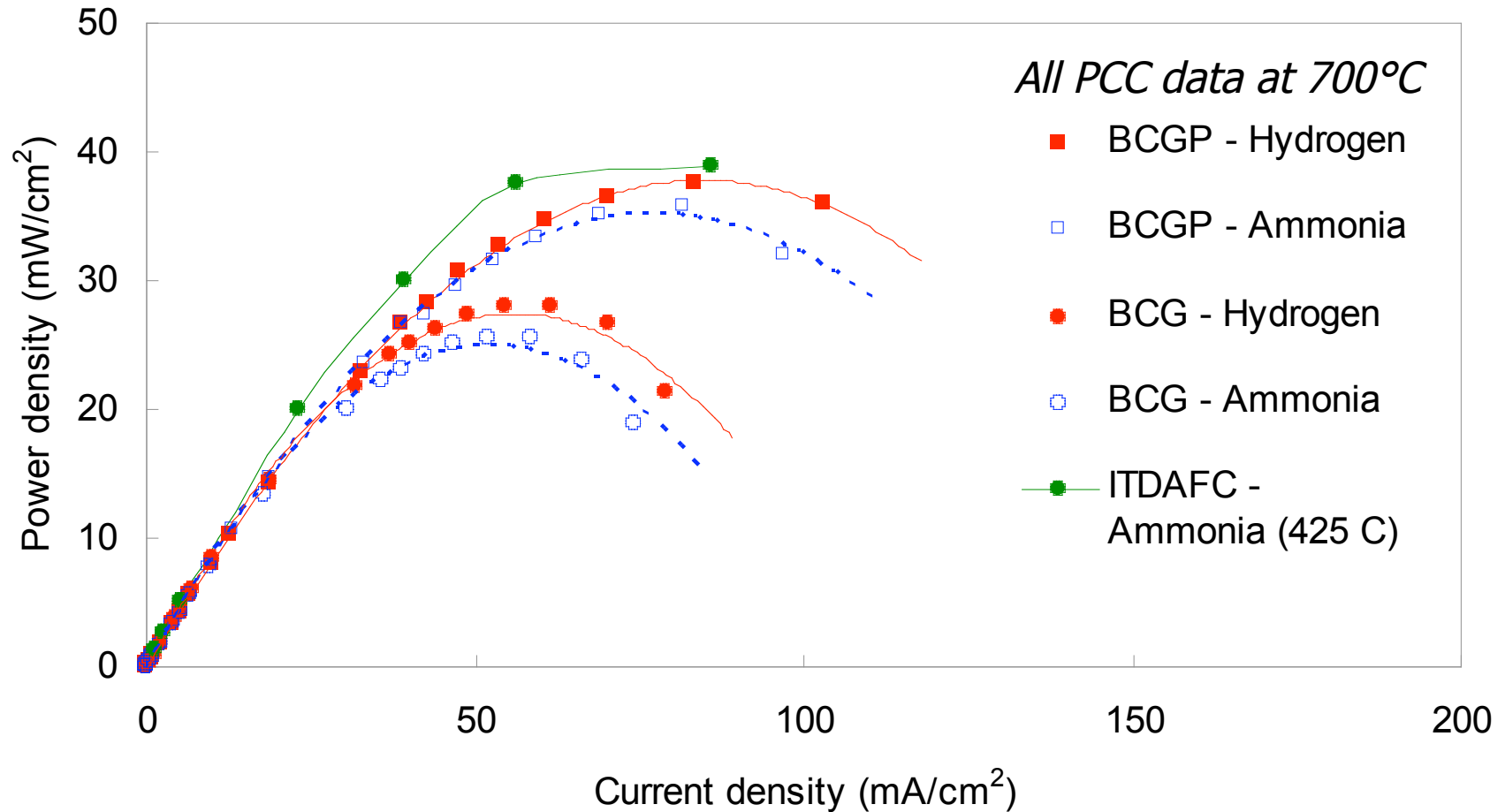


Ammonia Cell Performance

Ammonia Fuel



Comparison with NH_3 PCC



Modified from A. McFarlan, NRC



What's the plan? Build a Car, of Course



- **A med-temperature fuel cell main power plant: 90 kW**

- Ammonia fuel
 - ◆ Non-explosive, energy dense
 - ◆ Reactive-walled fuel tank
- 60 kW electric motor drive
 - ◆ Capable of delivering 300 kW bursts
 - ◆ Torque controlled electronically
 - ◆ Out-accelerates comparable ICE



The Holbrookmobile

- **Ultracapacitor bank**

- Regenerative braking
- Charges during cruising and "idling"
- Power dump to motor for bursts

EV-1 Conversion

- **General Motors donation to Howard University**
 - All technical manuals and specifications available
 - No battery pack
- **Goal: convert EV-1 to a hybrid fuel cell vehicle**
 - Challenges: FC design, installation, and control
 - Creation of a new project for undergraduate and graduate hands-on education
 - Will involve multiple disciplines



(General Motors)



Questions/Discussion



Jason C. Ganley
Howard University
Department of Chemical Engineering
2300 6th Street, NW
Washington, DC 20059
(202) 806-4796
jganley@howard.edu