



Renewable NH₃ and Direct NH₃ Fuel Cells: Canadian R&D for Clean Distributed Electricity Generation

**Presented at
9th Annual NH₃ Fuel Conference
San Antonio, TX**

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CanmetENERGY is the R&D branch of Natural Resources Canada, with headquarters and labs in Ottawa ON, and labs in Varennes QC and Devon AB



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Introduction

Anhydrous ammonia (NH_3), the chemical building block for nitrogen fertilizers, is also a carbon-free fuel.

Considerable infrastructure is in place to transport and store NH_3 due to its extensive use in agriculture.

Direct NH_3 fuel cells can potentially produce electric power and high grade heat (CHP) efficiently and with zero emissions

Like hydrogen, NH_3 can be produced from non-fossil renewable electricity sources (hydro, wind). NH_3 production from renewable electricity, and NH_3 as a renewable fuel is gaining attention worldwide, particularly in the Corn Belt of US Midwest.

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Topics:

- Conventional NH₃ production – energy requirements and GHG emissions
- Direct NH₃ fuel cell development at CanmetENERGY
- Direct fuelling of conventional SOFC's with NH₃
- NH₃ as a carbon-free renewable fuel
- R&D needs and opportunities

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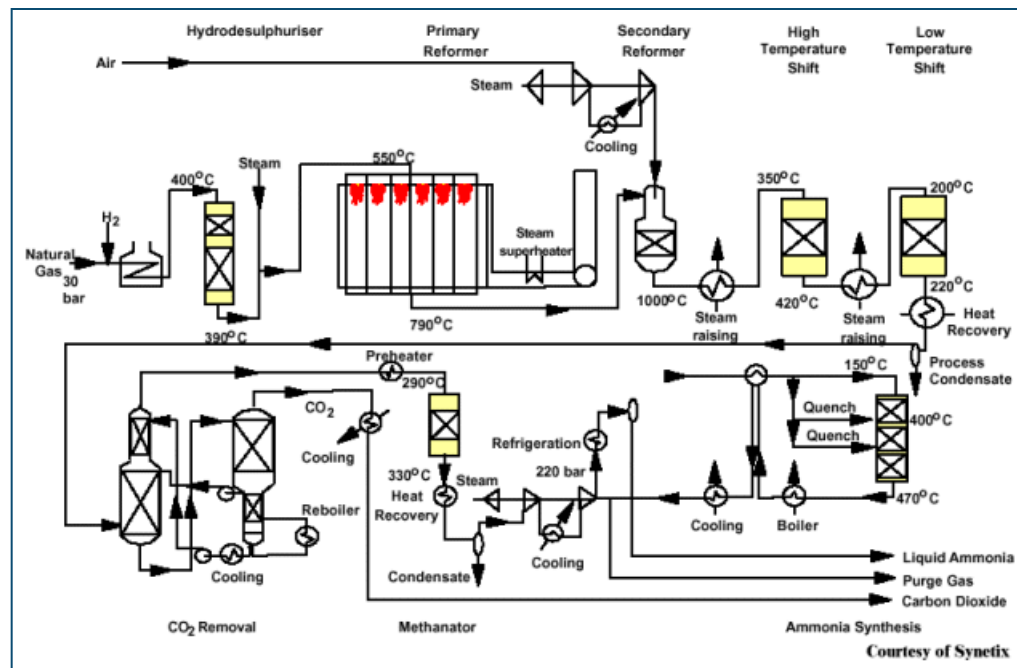
Conventional (Fossil) NH₃ Production

- ~200 million tonnes global capacity annually
- Production: 67% natural gas-based, 27% coal-based
- Use: >80% NH₃ is used for fertilizer manufacture
- US imports >50% of its NH₃
- Canada exports ~50% of its NG to US
- Also, NH₃ Production from NG or coal produces pure CO₂ byproduct which lowers cost of large scale CCS – and can be used for enhanced recovery of oil or coal-bed methane (NH₃ from coal with CCS practiced commercially in Beulah N.Dakota since 2000)

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NH3 Production from Natural Gas Reforming



Energy input: 30 million scf NG per tonne NH3
(30 GJ/tonne NH3)

CO2 Emissions: 1.8 tonnes CO2 per tonne NH3

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Overall Efficiency and CO₂ Emissions During Production and Distribution of Hydrogen Energy Carriers

(H. Anderson, World Hydrogen Energy Conference, Montreal, 2002)

Conclusions drawn from studies done by Norsk Hydro:

- CO₂ capture and sequestration contributes only slightly to the losses in the full hydrogen value chain
- **Central hydrogen and ammonia production seem to be the most efficient way to produce CO₂-free energy carriers**
- **Ammonia infrastructure development is easier because truck transport is possible** – supply and demand will be in balance through time
- On site natural gas reforming and methanol steam reforming have highest CO₂ emissions

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Direct NH₃ Fuel Cells:

Development of Proton-
Conducting Ceramic FC Materials
at CanmetENERGY

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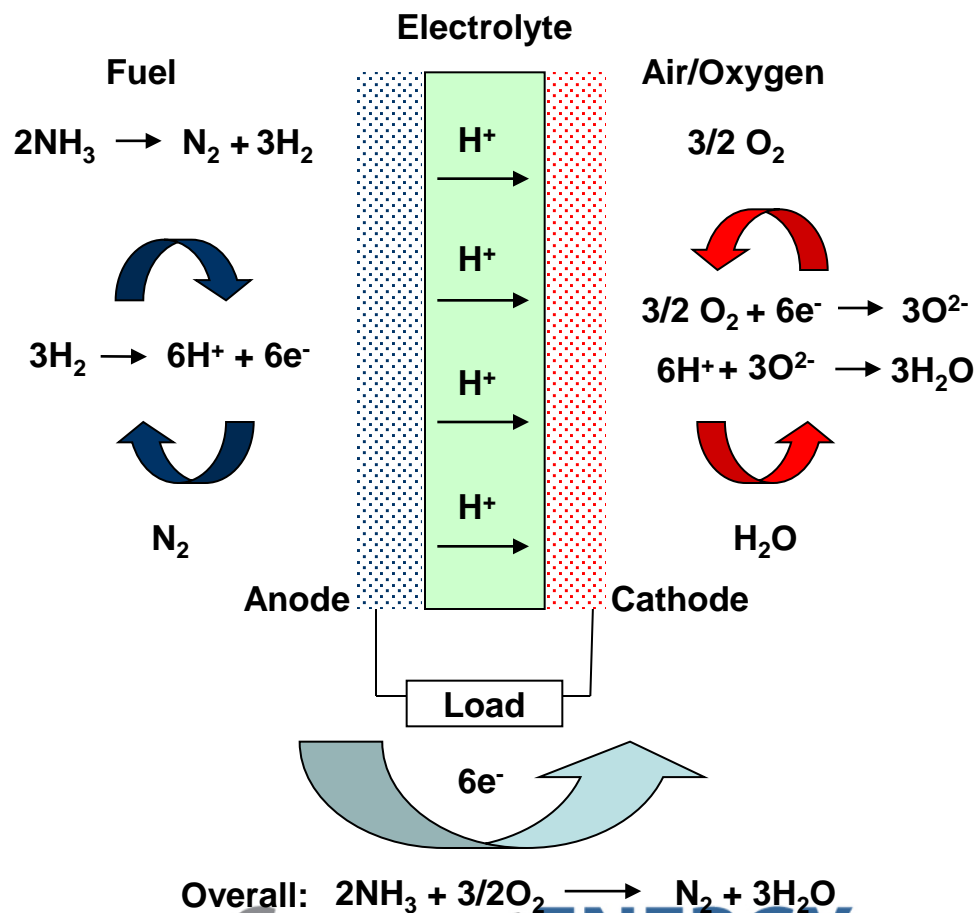
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Direct NH3 Fuel Cells:

What is the concept?

- Ammonia is catalytically decomposed to $\text{N}_2 + \text{H}_2$ at anode
- high temperature, low pressure favors equilibrium limited decomposition
- Protons transport across a solid proton conducting electrolyte.
- Removal of hydrogen at the anode drives decomposition reaction to completion.
- H_2 /air oxidation at the cathode provides chemical driving force for the fuel cell AND provides the heat of reaction for ammonia decomposition.
- Products of the fuel cell are nitrogen, water, electric power and heat.



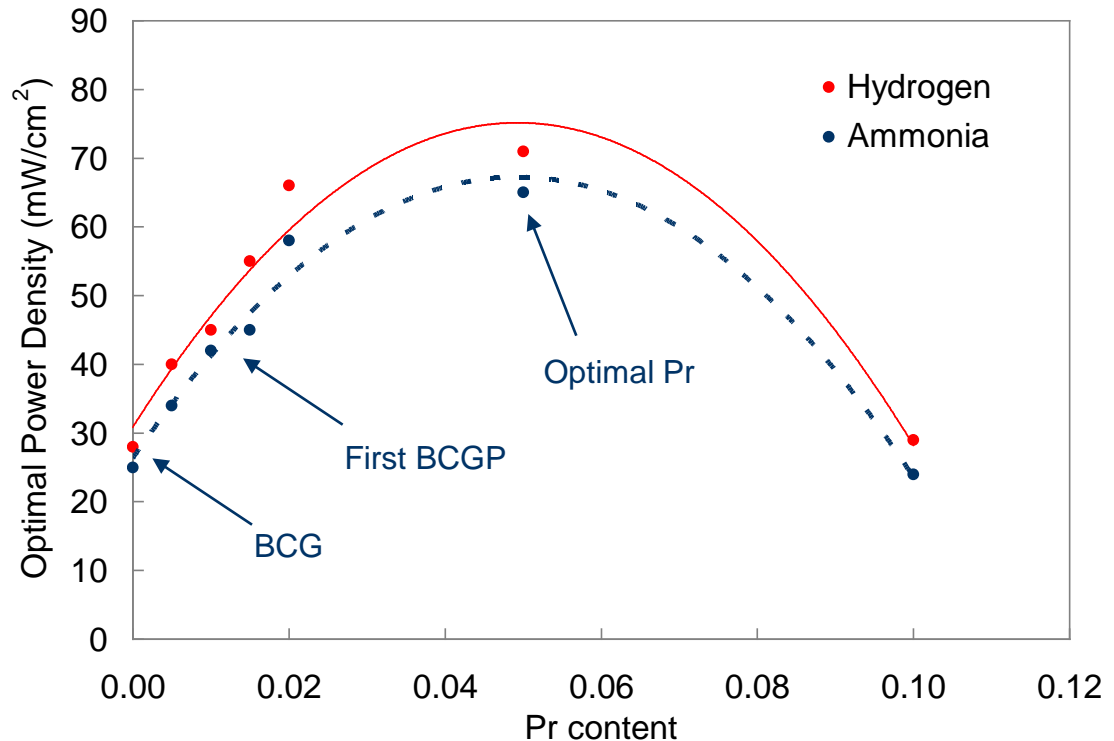
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Direct NH3 Fuel Cells:

Pt/BCGP/Pt Ammonia Single Cell Fuel Cell

- Cell performance as a function of Pr concentration (700° C).
- Optimization at approximately Pr = 0.05 ($\text{BaCe}_{0.8}\text{Gd}_{0.15}\text{Pr}_{0.05}$).
- Related to increased material density and decrease of cell unit volume.



Fuel Cells, volume 7, issue 4 (2007) 323.

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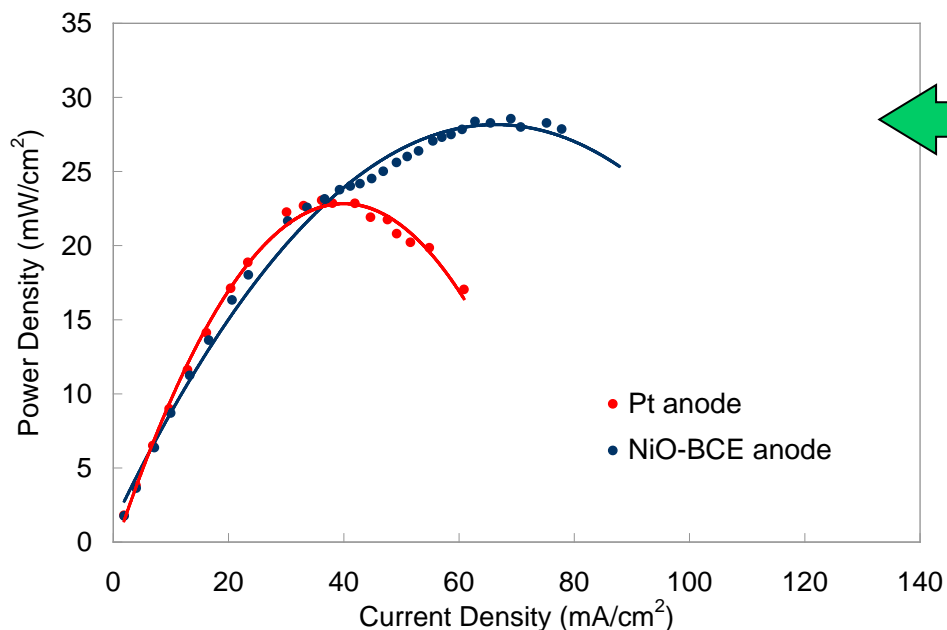
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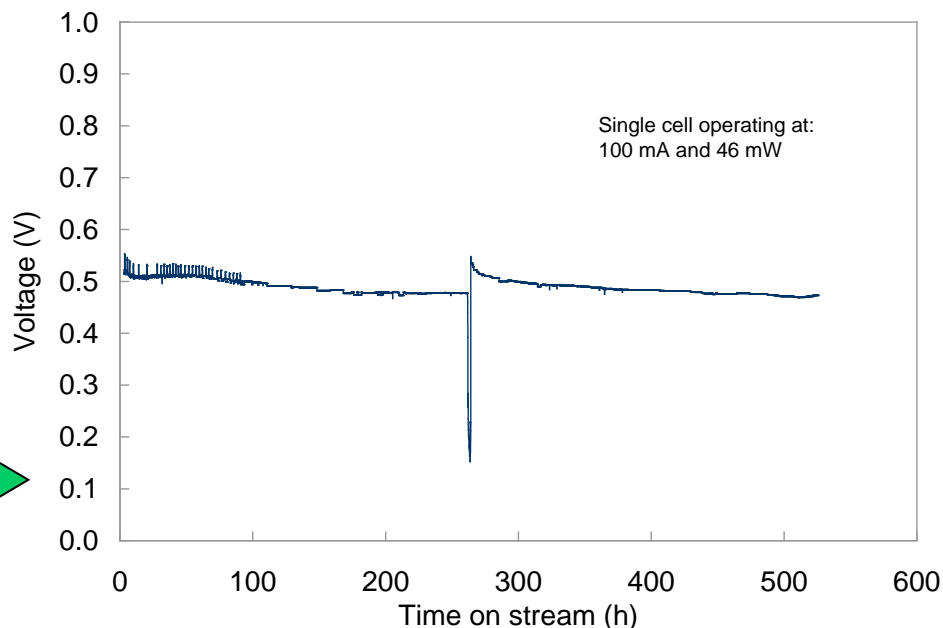
Direct NH₃ Fuel Cells:

Mixed ionic and electronic conducting anode



NiO-BCE anode outperformed Pt anode in a BCGP electrolyte supported fuel cell at 600° C in NH₃

Cell was stable for over 500 h on ammonia fuel with very little degradation in performance.



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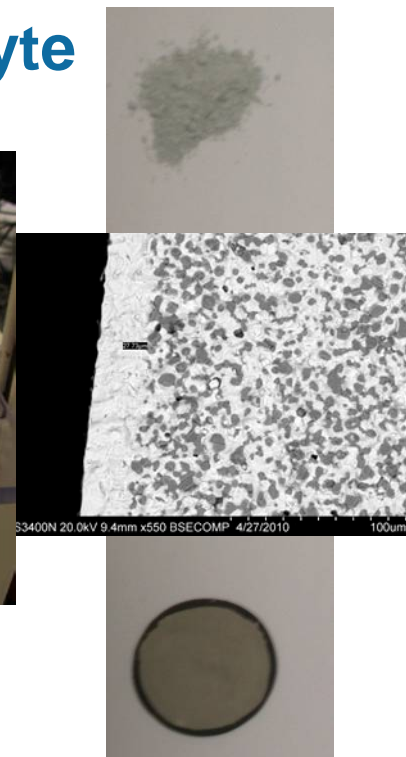
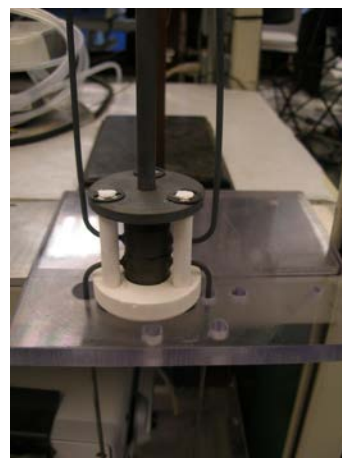
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Direct NH3 Fuel Cells

In-house development of advanced proton conducting ceramic fuel cell electrode/electrolyte materials:

- Proton conducting ceramics allow lower operating temperatures in H₂ and NH₃
- They are thermodynamically more efficient than oxygen ion conducting ceramics
- X-section of Samarium doped barium cerate (BCS) under scanning electron microscope shows a uniform and dense ~25 micron thickness electrolyte layer, supported by a porous 1 mm thickness anode composed of NiO-BCS cermet
- This single fuel cell exhibits correct material properties needed for producing a fuel cell stack.



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Direct NH₃ Fuel Cells:

Testing NH₃ on Conventional
Precommercial SOFC Stacks at
Acumentrics Canada

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Acumentrics Canadian Division was established in 2007 at the Fuel Cell Research Centre in Kingston ON. CanmetENERGY partnered with them to do field testing of their stack using direct NH₃ fuel.

RP-20 500W SOFC
systems at a well head
in Texas (2012)

Source:
www.acumentrics.com



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Direct NH3 Fuel Cells:

Acumentrics SOFC Direct NH3 Field Test

Objectives:

- Assess the technical feasibility of using ammonia as a direct fuel source in a SOFC micro CHP generator.
- Measure the long term stability and performance of the SOFC to see if the stack materials are degraded by direct NH3 fuel.
- Measure level of emissions, especially NOx in the stack effluent gas
- Determine the modifications required to a natural gas based system to allow it to operate on ammonia



Acumentrics SOFC stack:

- 9x5 array of extruded tubes
- NiO/YSZ anode support
- YSZ electrolyte
- shell (air) side cathode
- nominal 1 kW output

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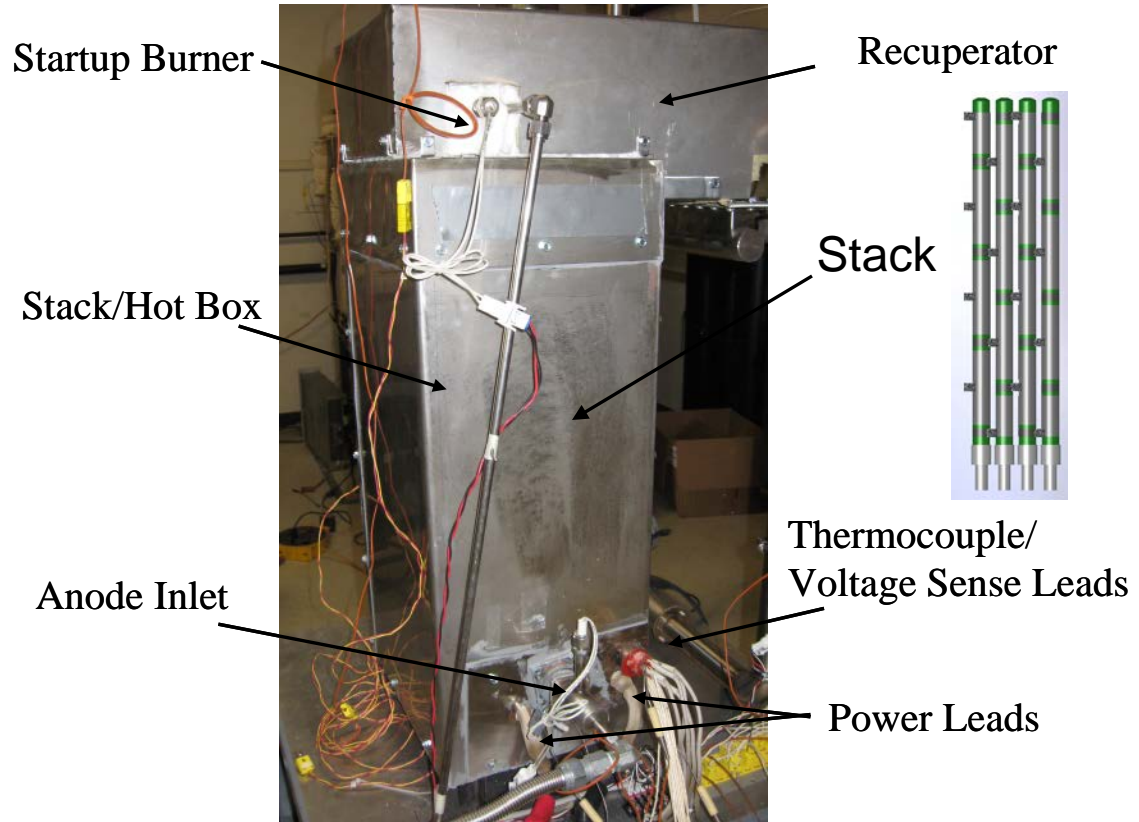
Direct NH₃ Fuel Cells:

Acumentrics Custom SOFC Test Stand



Test Stand

Fuel Cell Module



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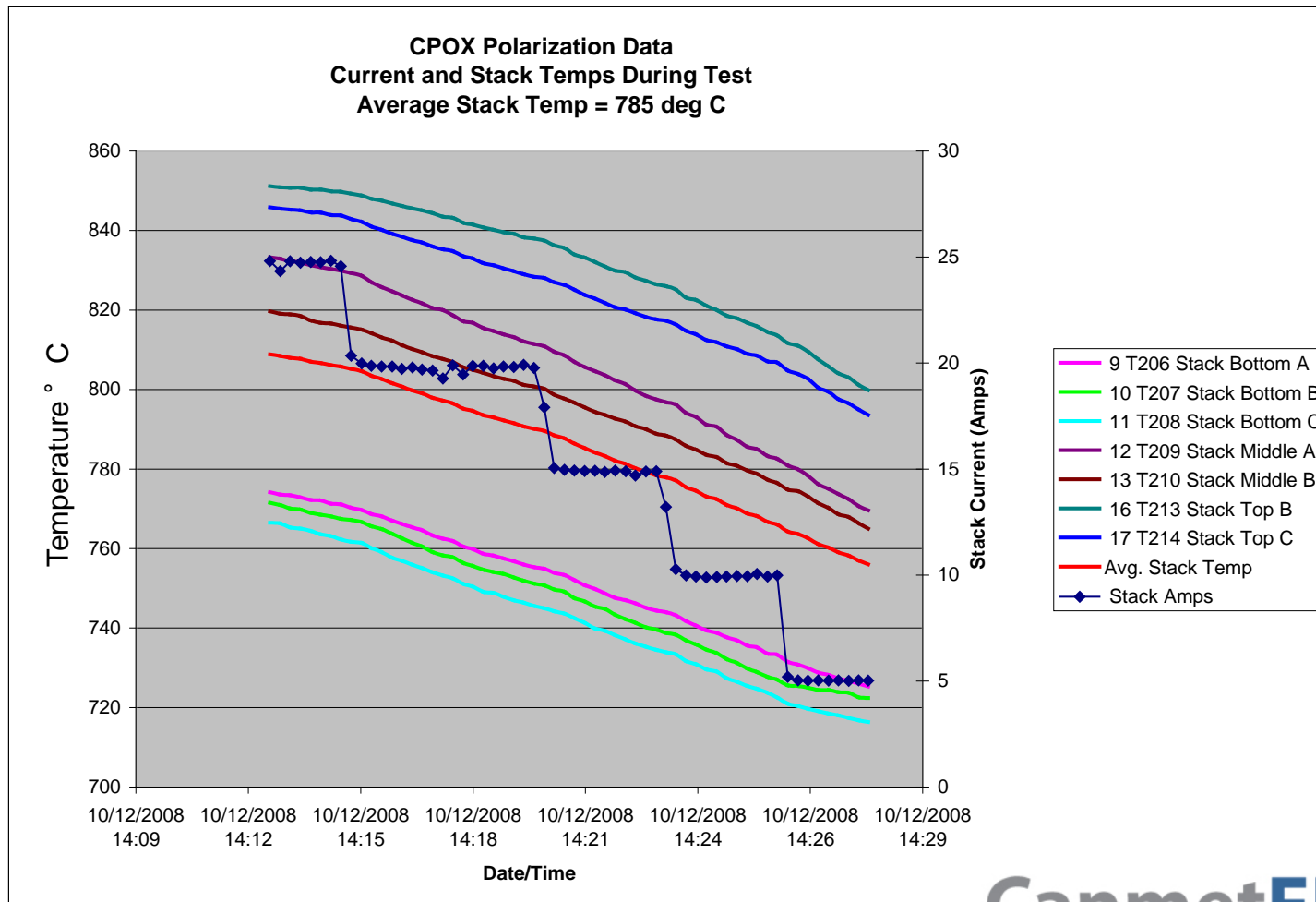


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Direct NH3 Fuel Cells: Acumentrics SOFC Stack Under Test



Temperature profile for CPOX Polarization Curve

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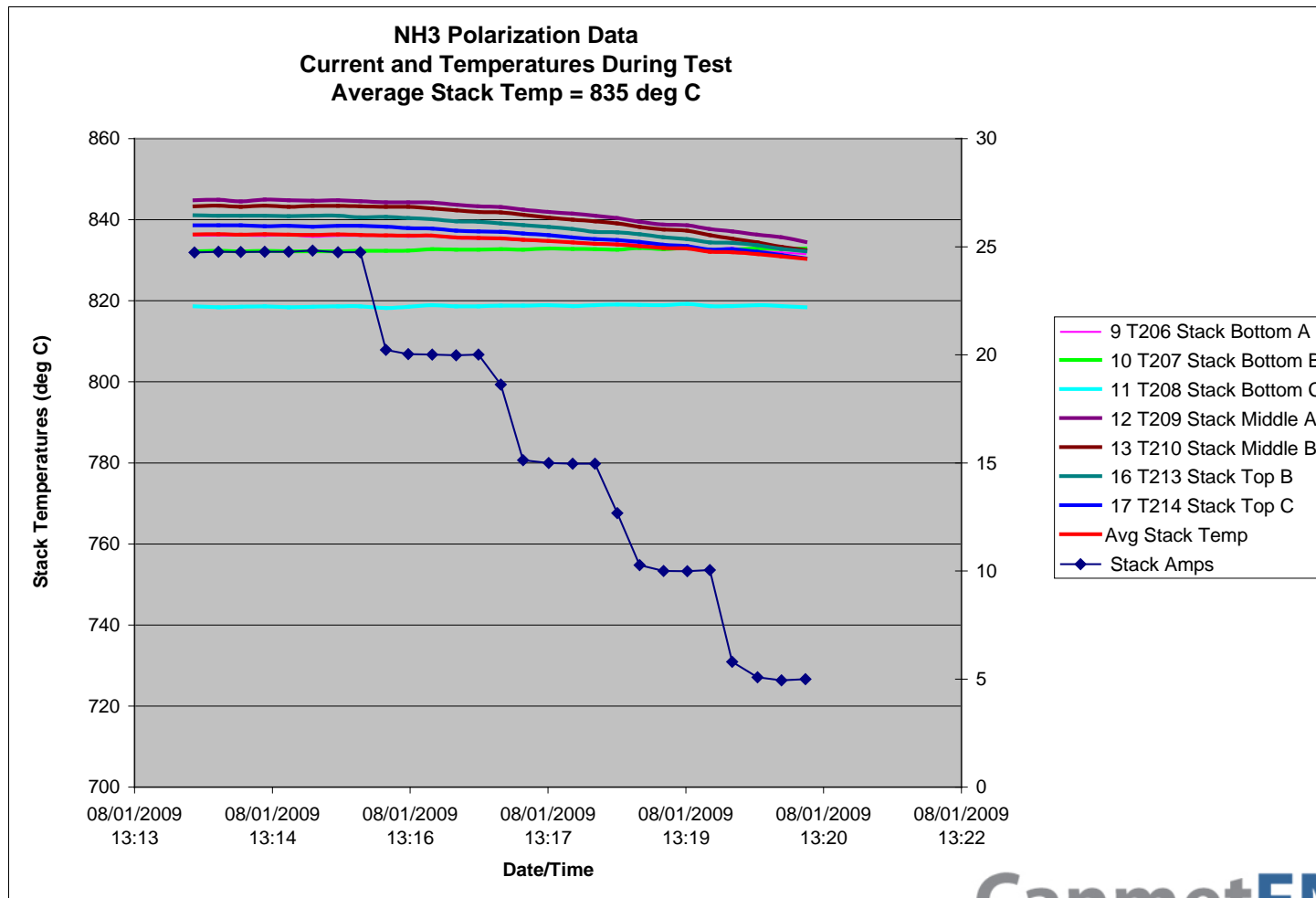


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Temperature profile for NH3 Polarization Curve

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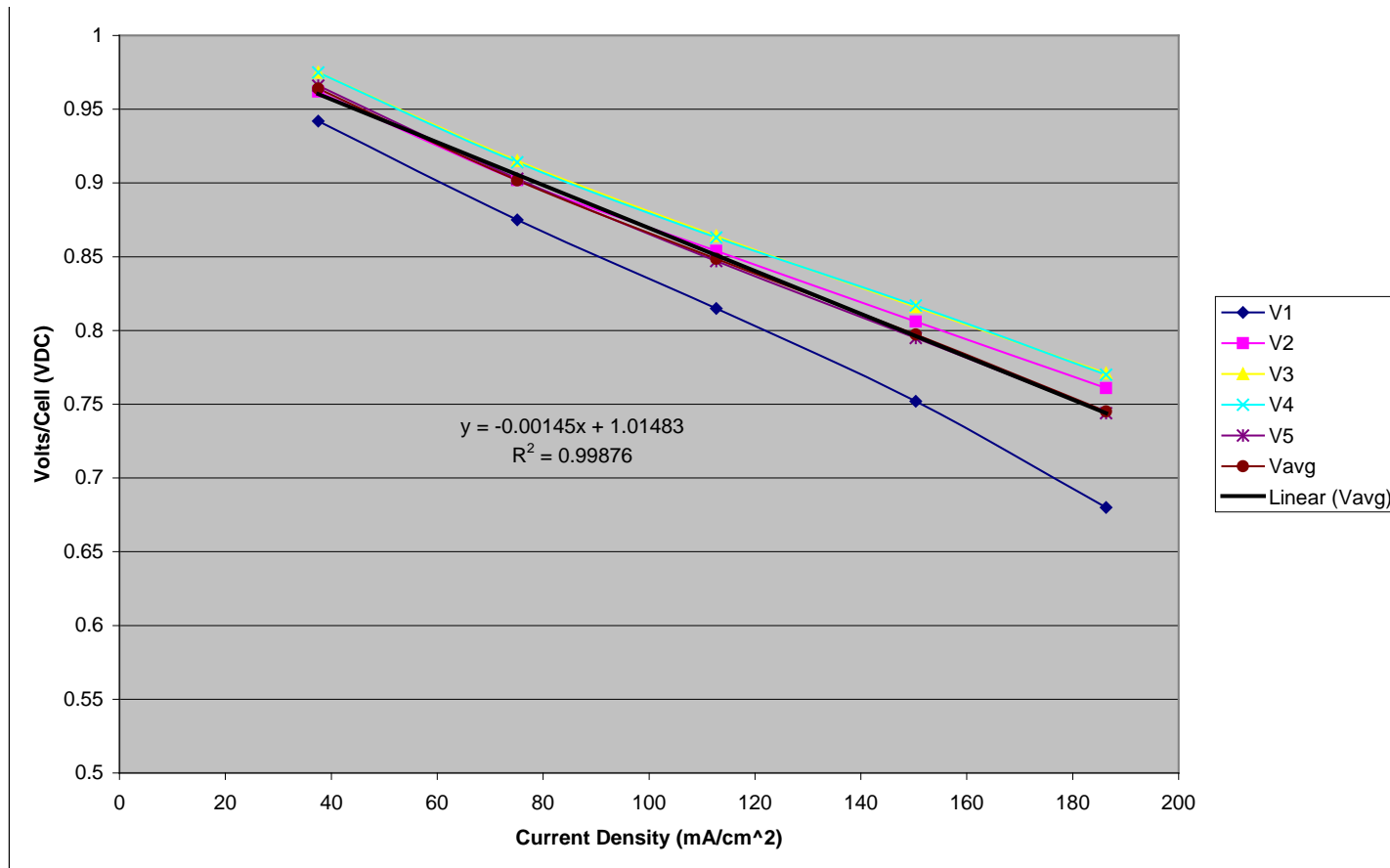


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NH3 Polarization Curve

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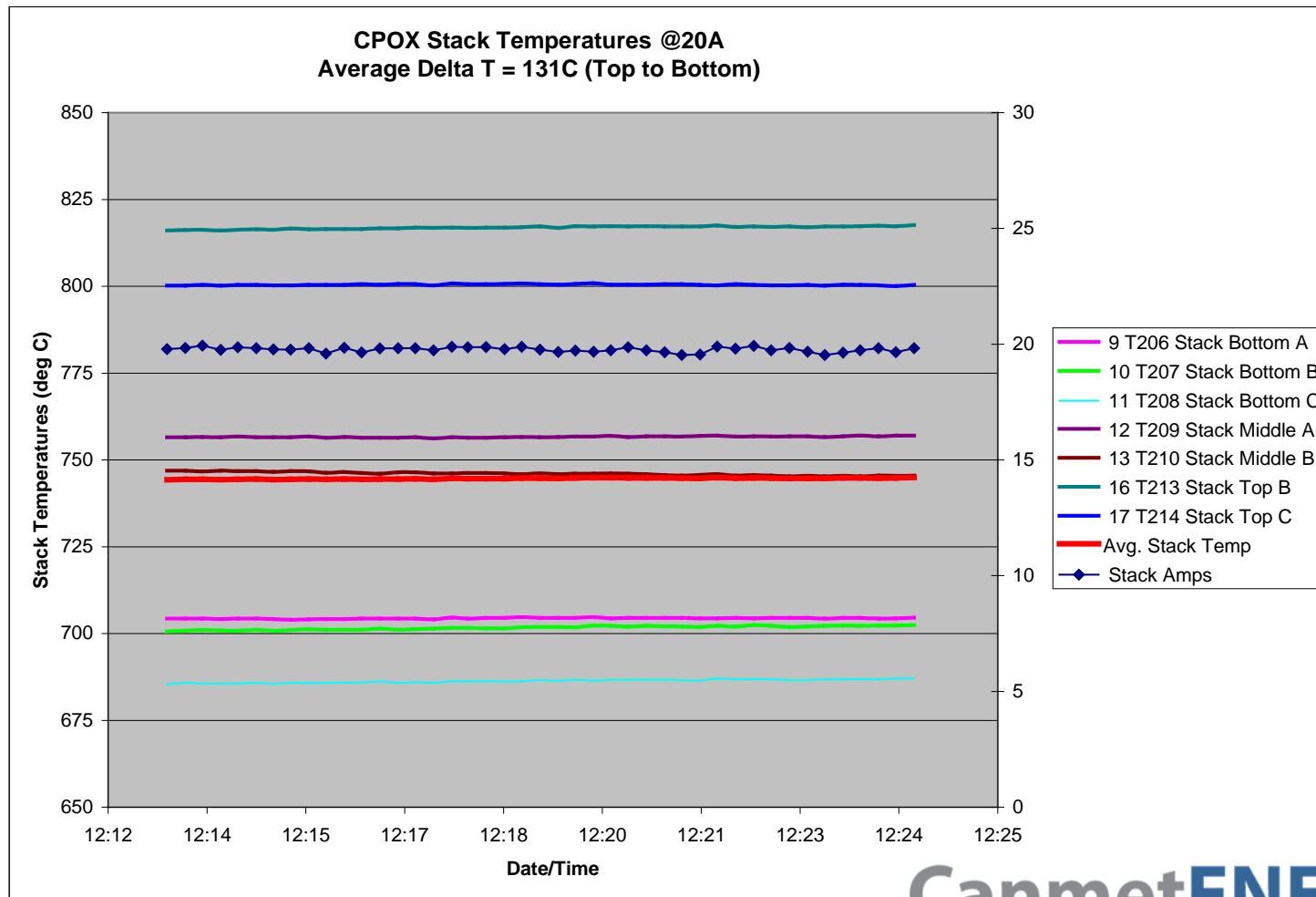


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CPOX (Natural Gas) Operation @20A

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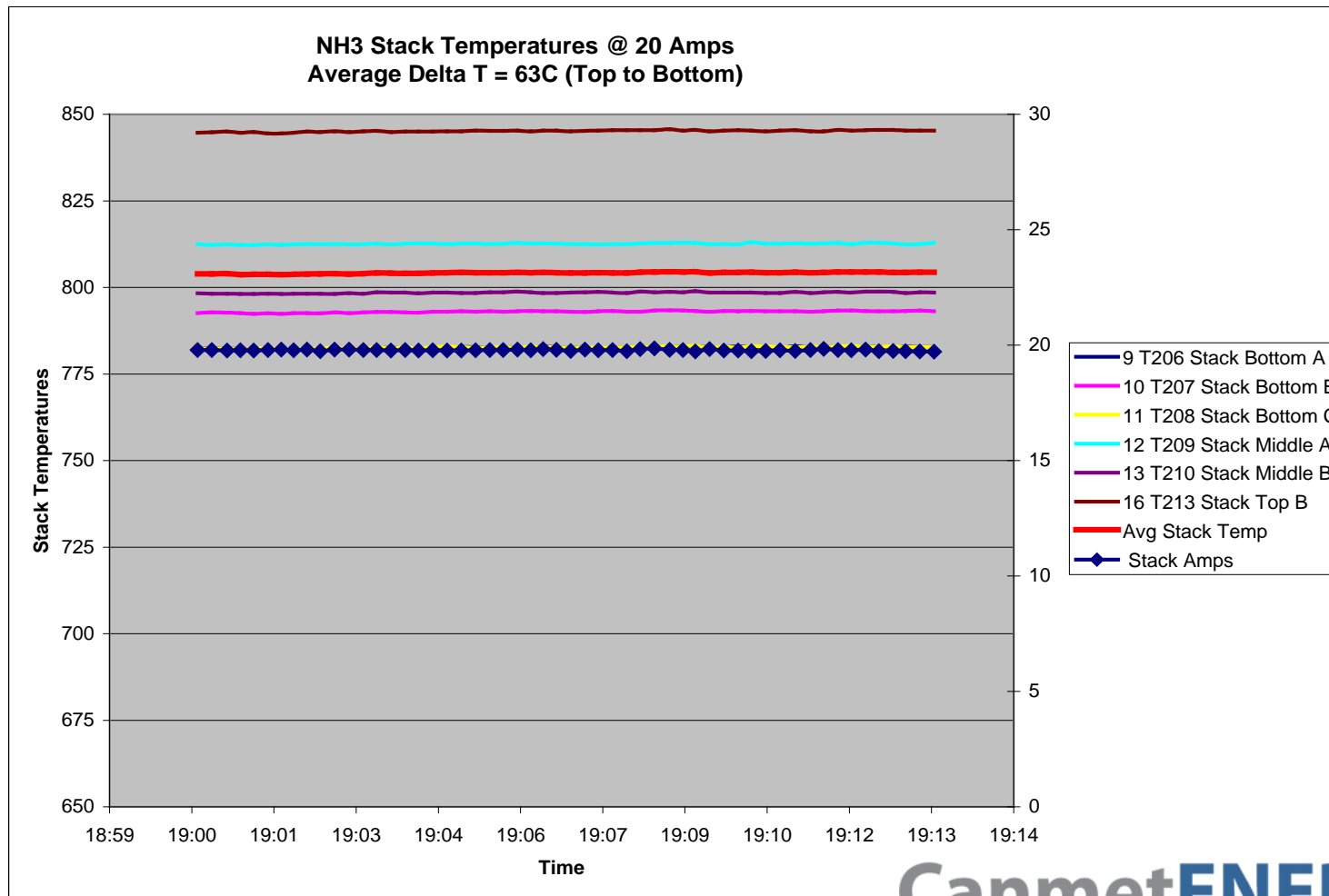


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Direct NH3 Fuel Cells: Acumentrics SOFC Stack Under Test



NH3 Operation @20A

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Direct NH₃ Fuel Cells:

Acumentrics NH₃ SOFC Field Test

Knowledge Gained:

- Long term operation (~1400h) showed no adverse effects or drop off in stack performance for conventional SOFC.
- No residual NH₃, or NO_x was detected in the stack effluent gas, or effectively zero emissions operating on ammonia.
- NH₃ performance surpasses natural gas due to lower stack temperature gradient
- There is potential to simplify the system design considerably by using NH₃, lowering cost to build the system

Independently in 2010, Topsoe Fuel Cells (Denmark) presented similar findings using their near commercial SOFC's



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Production and Use of Renewable “Green” NH₃

NextHydrogen Study

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NEXTHydrogen is a Canadian clean technology company who is developing MW scale electrolyzers as well as “green” applications of this technology

NEXT HYDROGEN

Economical Clean Hydrogen, at Scale



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Canadian Opportunity for “Green” NH₃: NextHydrogen Study



- Renewable NH₃ (by water electrolysis, air separation, Haber Bosch synthesis) from low cost hydroelectricity (\$0.02/kWh - \$0.35/kWh) can be competitive with fossil NH₃ in the near future. Windfarms with low installed costs and good to very good wind availability may also compete.
- Renewable (green) NH₃ addresses energy security – for both energy and food production by reducing the industry’s dependency on Canada’s NG, heavy oil and coal resources.
- Green NH₃ is an emerging decentralized (off grid) energy resource. It has the potential to reduce the GHG emissions related to fossil NH₃ production.
- Demonstration ready technologies for using green NH₃ as a fuel are available today. They include NH₃ in internal combustion engines, and solid oxide fuel cell combined heat and power systems. NH₃ production technology based on electrolysis is also at demo stage of development.

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Economics of Natural Gas and Coal Based NH₃ Production

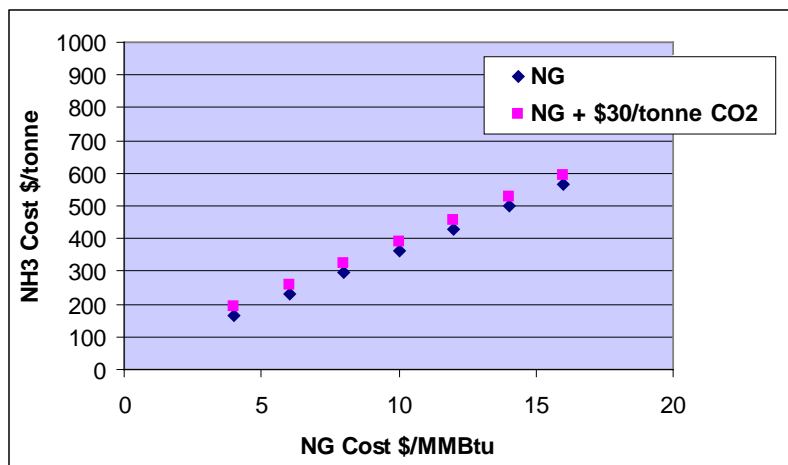


Fig.1

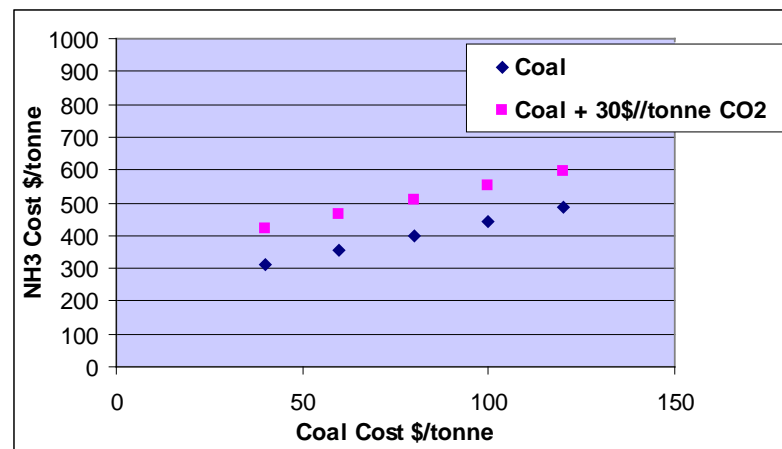


Fig.2

Fig.1 and Fig. 2 show that low-cost NG and/or coal based NH₃ production remain economical for large scale production. Carbon Capture adds incrementally to production cost, more so for coal than NG.

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Economics of “Green NH3”

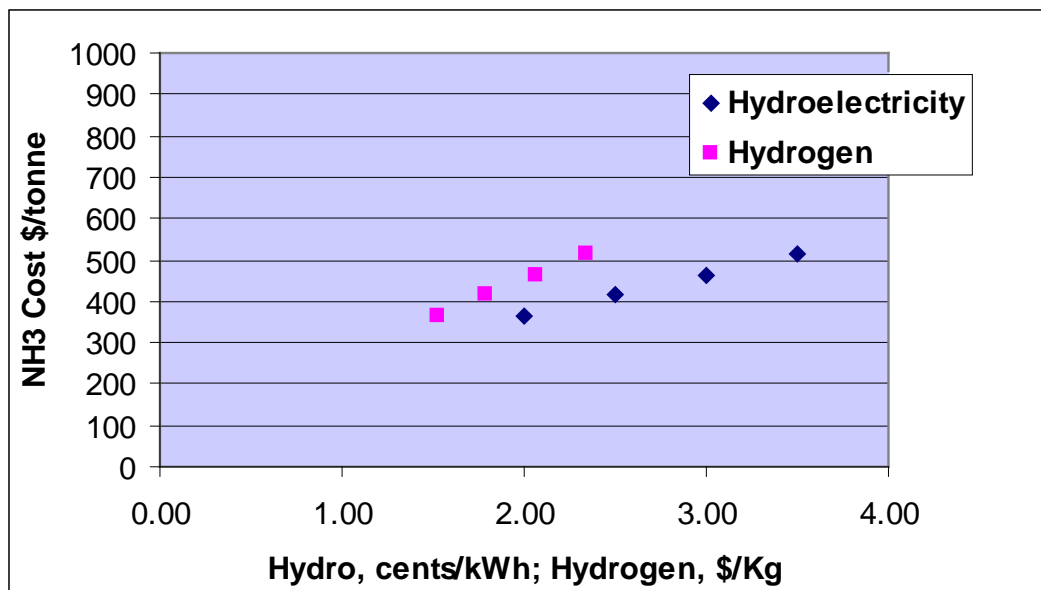


Fig.3

Fig. 3 shows that renewable NH3 from low cost hydroelectricity (\$0.02-\$0.35 /kWh) can be competitive with fossil NH3 in the near future. Hydrogen costing \$1.50-\$2.50 /Kg from wind energy will also be competitive.

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Economics of “Green NH3”

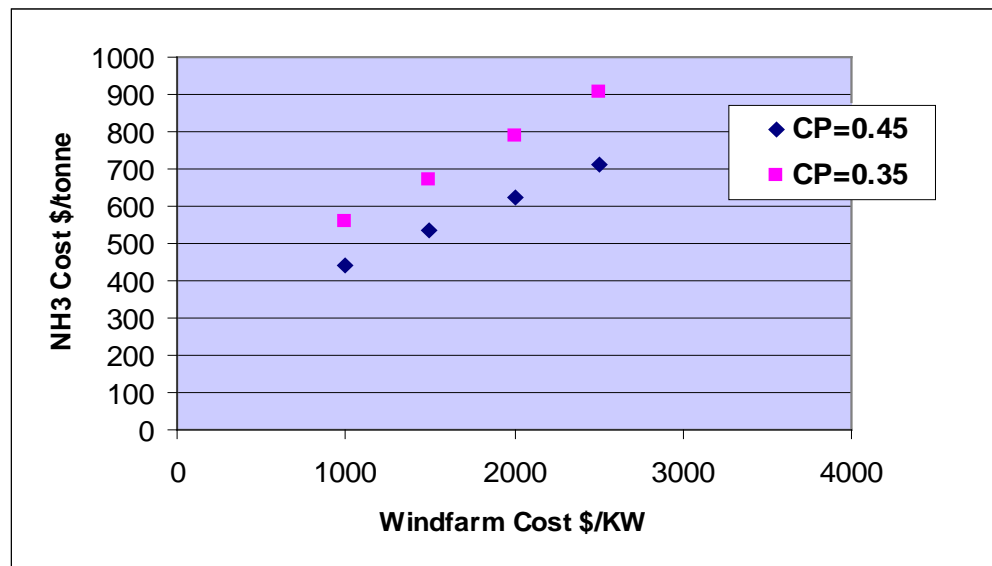


Fig.4

Fig. 4 shows the strong influence of wind farm installed cost and wind capacity factor (CP) on renewable NH3 cost. Wind farms require an installed cost approaching ~1000/kW, good wind (CP=0.35) or even very good wind (CP=0.45) to be competitive in today's NH3 market.

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Next Steps



- Pursue renewable NH_3 production and technology demonstration tailored to match economic advantages of a given region.
- Pursue new applications for green NH_3 , e.g. farm vehicles, SOFC for CHP and refrigeration in regions where existing NH_3 infrastructure can be leveraged.
- Pursue R&D on proton conducting ceramic electrolyte materials for efficient solid state electrochemical synthesis of ammonia.
- Nurture emerging Canadian expertise and technology development in related areas, and strengthen linkages nationally and internationally with entities having similar interests.
- R&D to advance biomass gasification for green NH_3

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Potential for Green NH₃ Production From Biomass Gasification

ENERKEM (right) and **NEXTERRA** (below) are two Canadian companies who are developing syngas technologies based on biomass residues and MSW. These syngas processes could be utilized to produce liquid fuels, RNG, and green ammonia.



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“Green” NH₃ – An option for distributed clean energy and distributed NH₃ production in Canada ?

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