

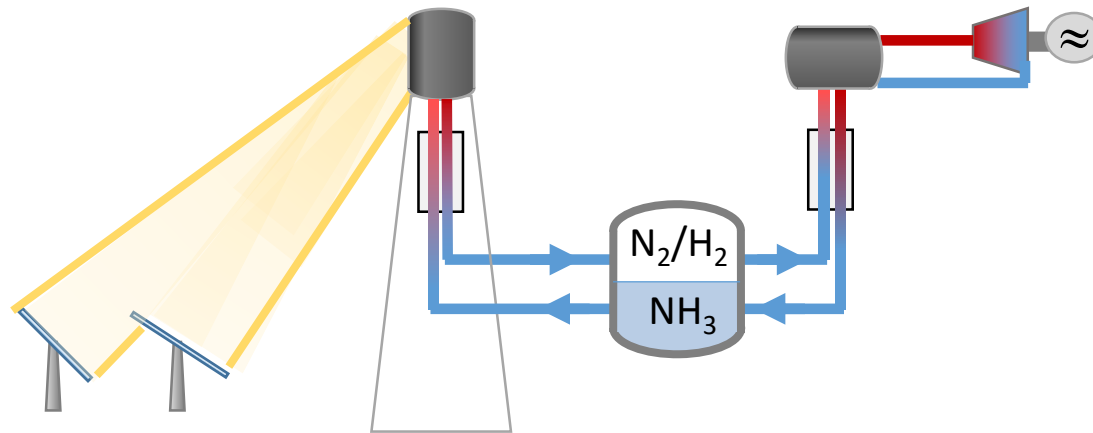
Thermochemical Energy Storage with Ammonia & Implications for Ammonia as a Fuel

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September 19, 2016

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Overview of talk

- Introduce technology
- Discuss findings
- Pose questions: How our research may be relevant to synthesizing ammonia as a fuel

INTRODUCTION AND BACKGROUND

Concentrating Solar Power (tower configuration shown)



Crescent Dunes

110 MW_e, 10 hrs molten salt
energy storage

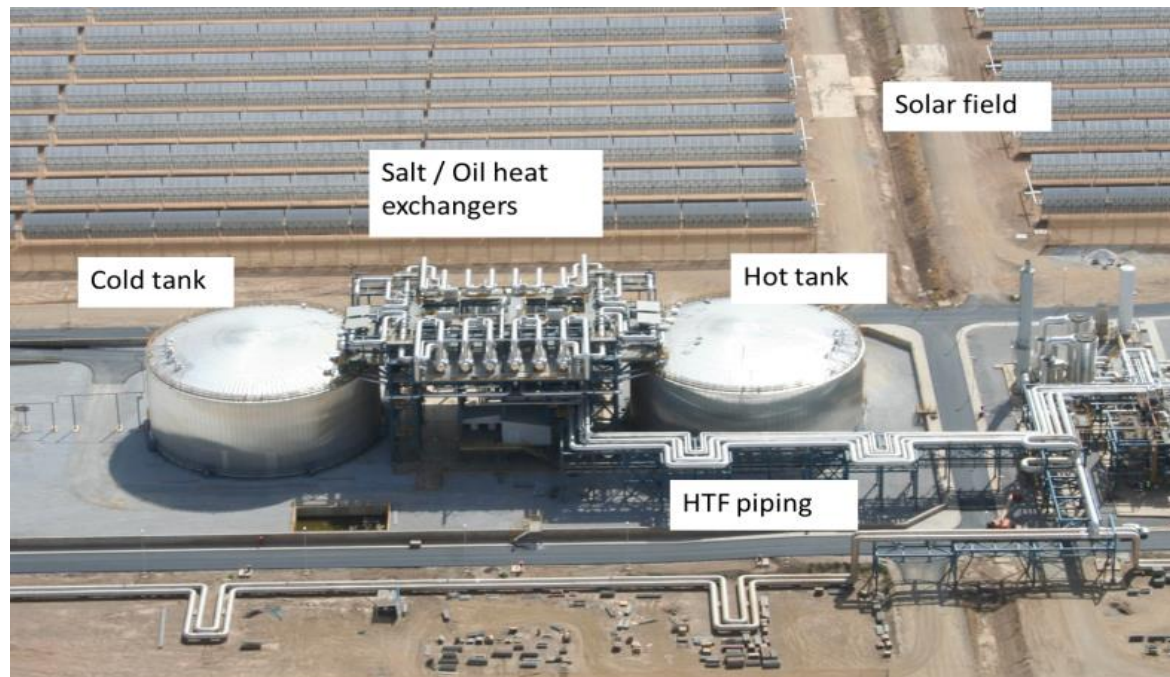


Ivanpah

400 MW_e, largest ever
CSP plant

Energy storage is CSP's competitive advantage

- Thermal storage enables electricity generation independent of time of day.
- Storage makes better use of the plant investment, can reduce LCOE.
- State of the art: two-tank molten salt storage.



Andasol 3
Courtesy
Ferrostaal

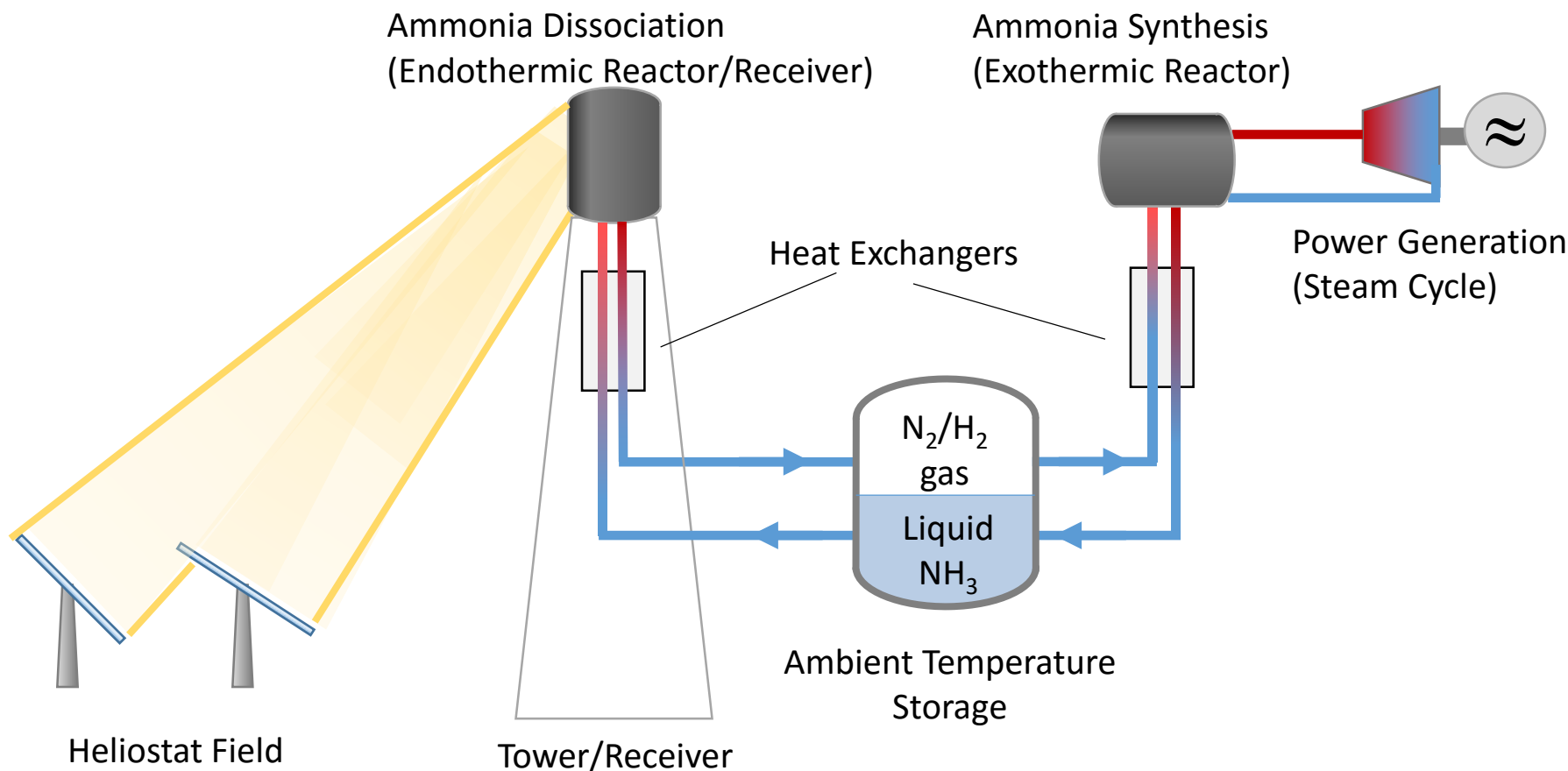




Salt itself is limiting cost factor
~ \$25-40/kWh of energy stored

- U.S. Dept. of Energy SunShot supports research into energy storage for CSP
- Performance Goal: Recover heat at 650°C to enable advanced power block
- Target for Capital Cost: \$15 per kWh of energy stored
 - not to be confused with LCOE
 - denominator not to be confused with energy for combustion of NH_3
- Many ideas, few are proven technologies
- Ammonia-based thermochemical energy storage has the potential to meet the performance and cost metrics

System overview



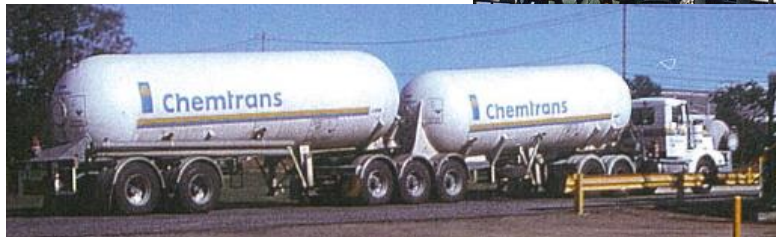
Pros and cons of ammonia TCES

Pros

- Extensive industrial experience
 - Haber Bosch
 - Catalysts available
 - Transportation routine
- Low cost medium
- Ambient temperature storage
- Automatic phase separation between products/reactants
- And more...

Cons

- Modest energy density
- Necessity of storing gaseous components
- High pressure process



Prior ANU research demonstrated complete loop



12 kW
reactor/
receiver

20 m² dish concentrator



10 liter storage vessel



Synthesis reactor

Achieved wall temperature of
475°C

We want 650°C...

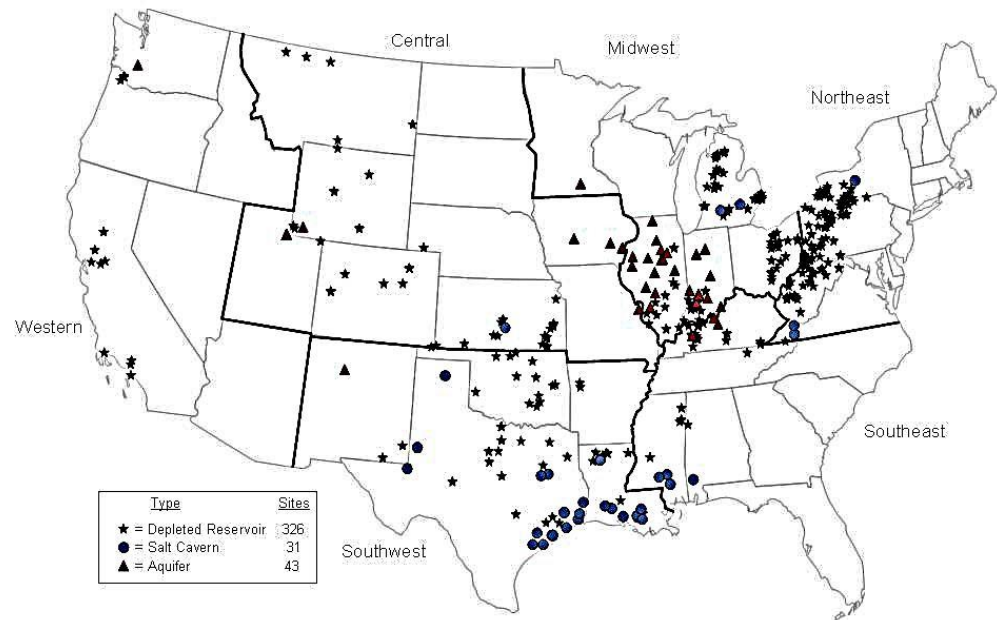
Key challenges identified and addressed

- Key challenges:
 - Can physical storage of high pressure nitrogen/hydrogen mixture be done cost-effectively?
 - Ammonia synthesis had never been used to heat supercritical steam to 650°C. Is it possible?
- Results presented today:
 - Gas storage
 - Heat recovery to supercritical steam at 650°C
 - Optimizing the synthesis reactor system

GAS STORAGE

Underground gas storage is prevalent

- Need to store ambient temperature, high pressure N_2+3H_2 .
- Underground storage concept:
 - Surrounding geology provides bulk of pressure containment.
- Approaches considered:
 - depleted oil or gas wells
 - aquifers
 - salt caverns
 - rock caverns
 - tunnel drilling
 - shaft drilling



Underground natural gas storage sites in US

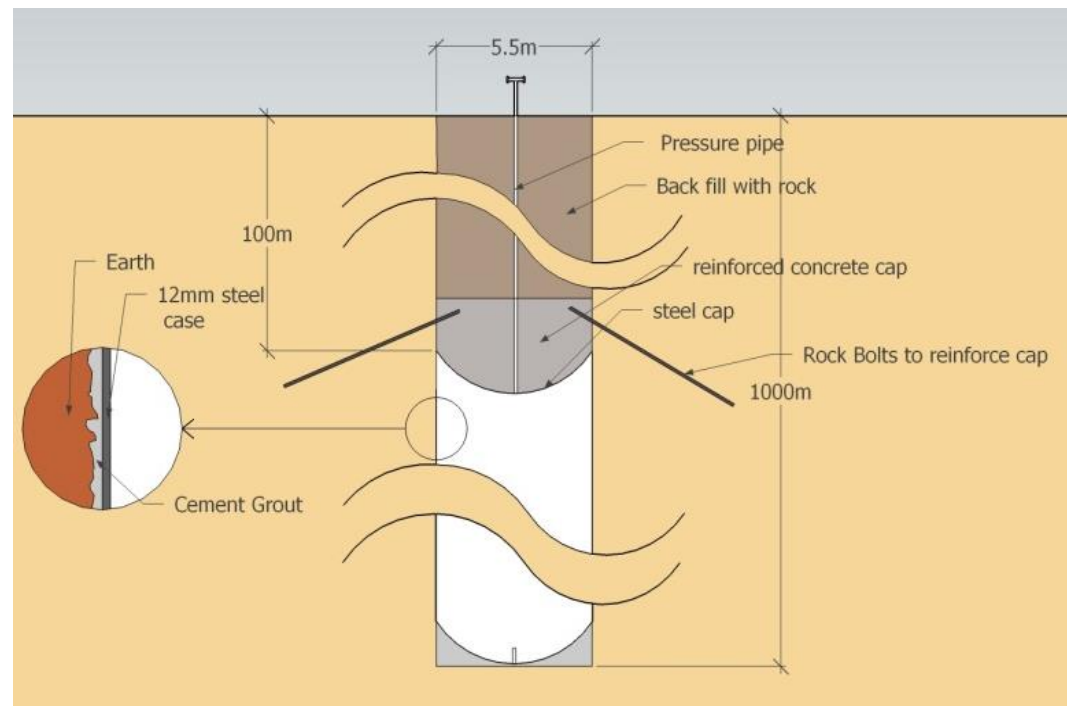
Salt caverns are inexpensive

- Solution mining of salt caverns is simple, established process
- Salt caverns widely used for storage:
 - Over 2000 salt caverns in North America alone for hydrocarbon storage.
 - Pure hydrogen or hydrogen-rich gas mixtures have been stored.
- Salt cavern conditions are suitable for our application:
 - Sufficient volume and pressure
 - Low permeability of rock salt
- Roughly \$1/kWh to create storage space (for large projects).
- Available on every continent
 - but does present a site constraint.



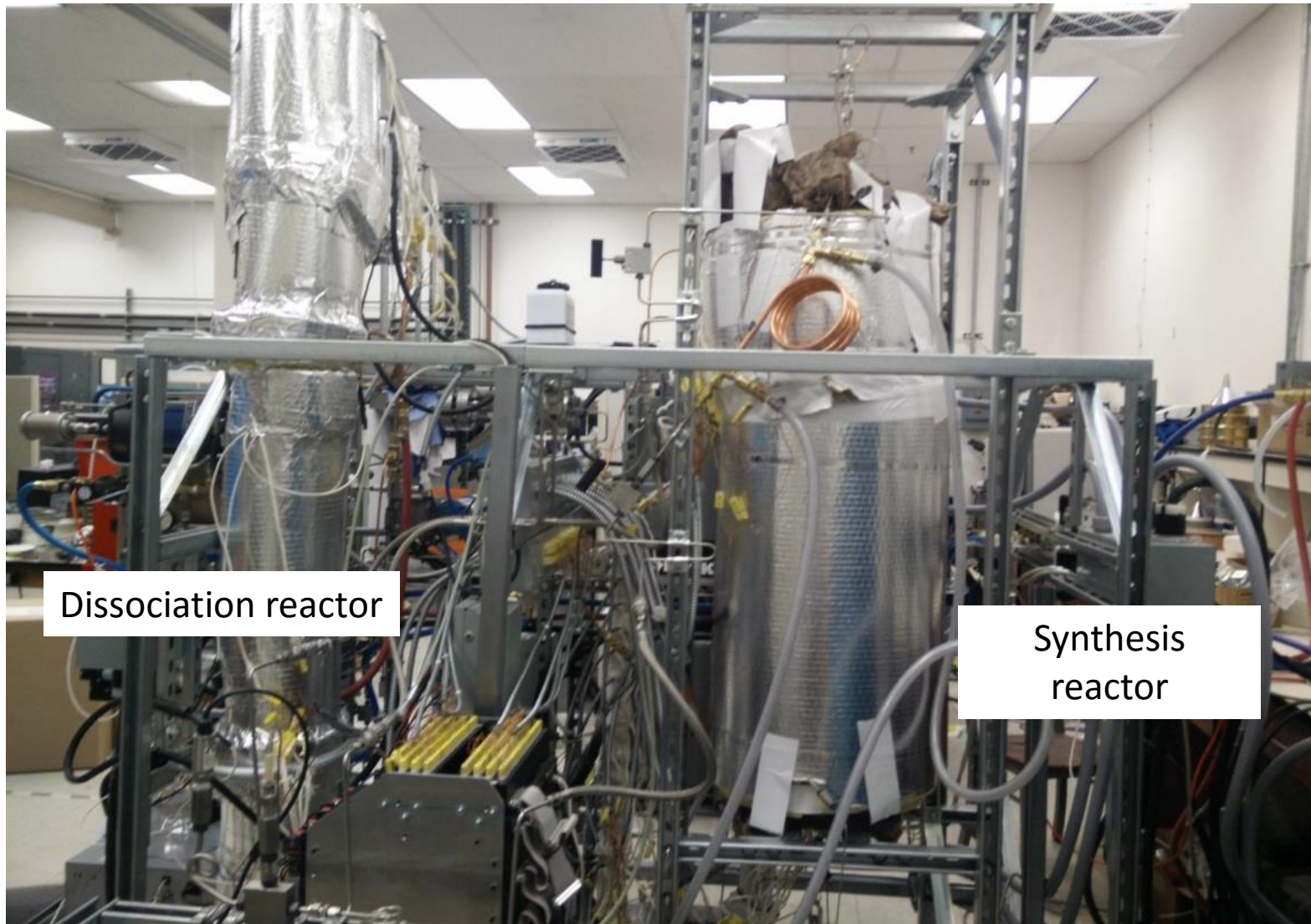
Large diameter drilled shafts provide another option

- Removes site choice constraint.
- Shaft drilling routinely carried out at up to 7.5 m diameter and depths of 1000 m.
- In consultation with drilling company:
 - Cost roughly \$5/kWh.
- Conceptual design developed.
- Details of hydrogen impermeable lining and endcaps required.



AMMONIA SYNTHESIS FOR HEAT RECOVERY TO SUPERCRITICAL STEAM

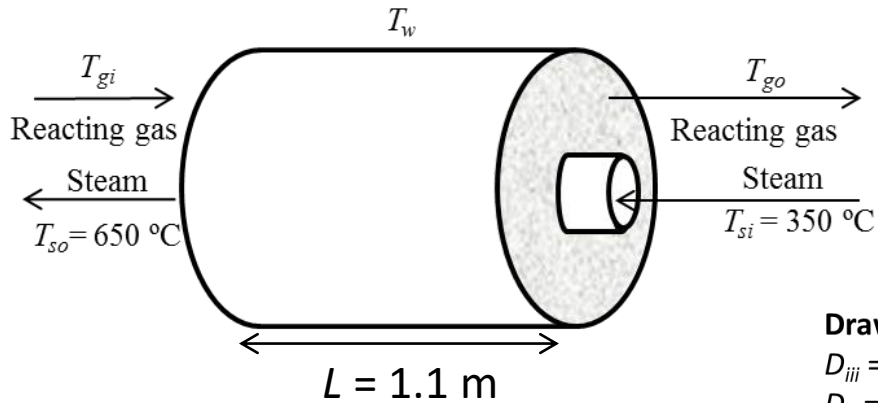
Experimental system in our lab



Dissociation reactor

Synthesis
reactor

Reactor-hx configuration



Drawn to scale.

$D_{iii} = 1.4\text{ mm}$

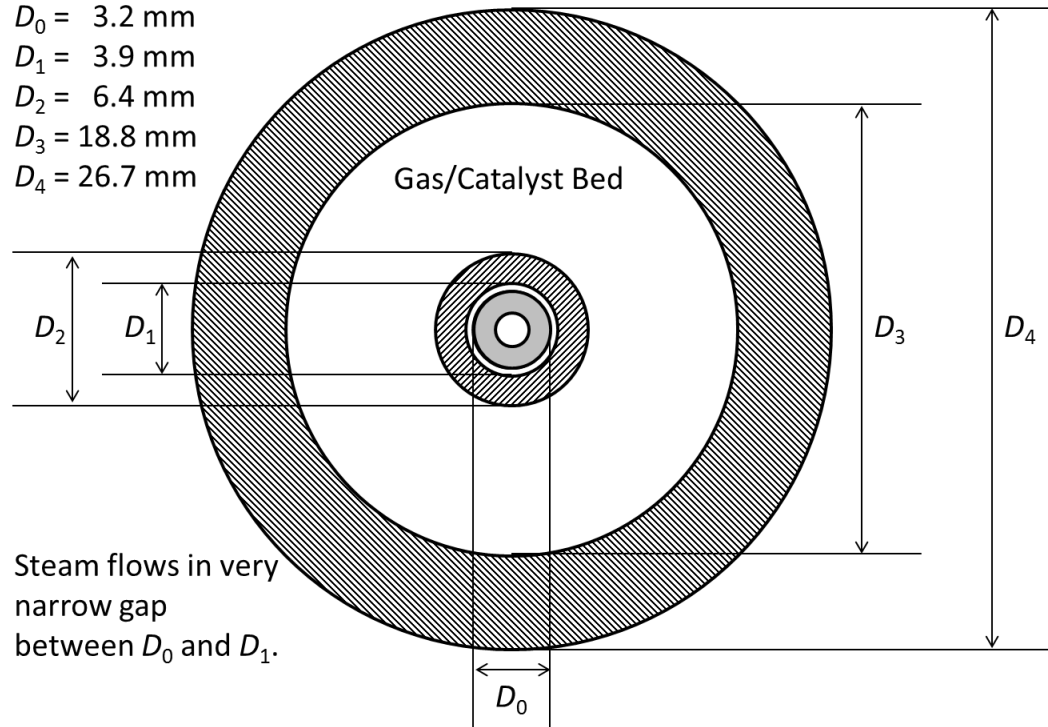
$D_0 = 3.2\text{ mm}$

$D_1 = 3.9\text{ mm}$

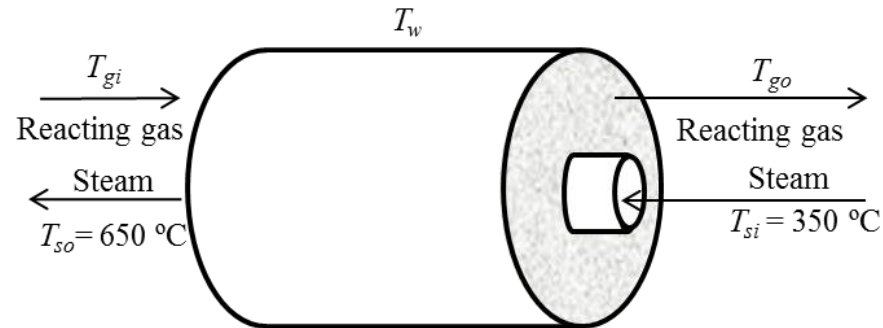
$D_2 = 6.4\text{ mm}$

$D_3 = 18.8\text{ mm}$

$D_4 = 26.7\text{ mm}$



Reactor-hx model



Gas in Catalyst Bed

(pseudo - homogeneous model)

$$\rho_g v_g C_{P,g} \frac{\partial T_g}{\partial x} = \frac{1}{r} \frac{\partial}{\partial r} \left(k_{eff} r \frac{\partial T_g}{\partial r} \right) + \dot{r}''' \Delta H$$

$$\rho_g v_g \frac{\partial f_{NH3}}{\partial x} = \rho_g D_{eff} \frac{1}{r} \frac{\partial}{\partial r} \left(r \frac{\partial f_{NH3}}{\partial r} \right) + \dot{r}'''$$

k_{eff} from Argo and Smith, 1953

Temkin - Pyzhev Rate Equation

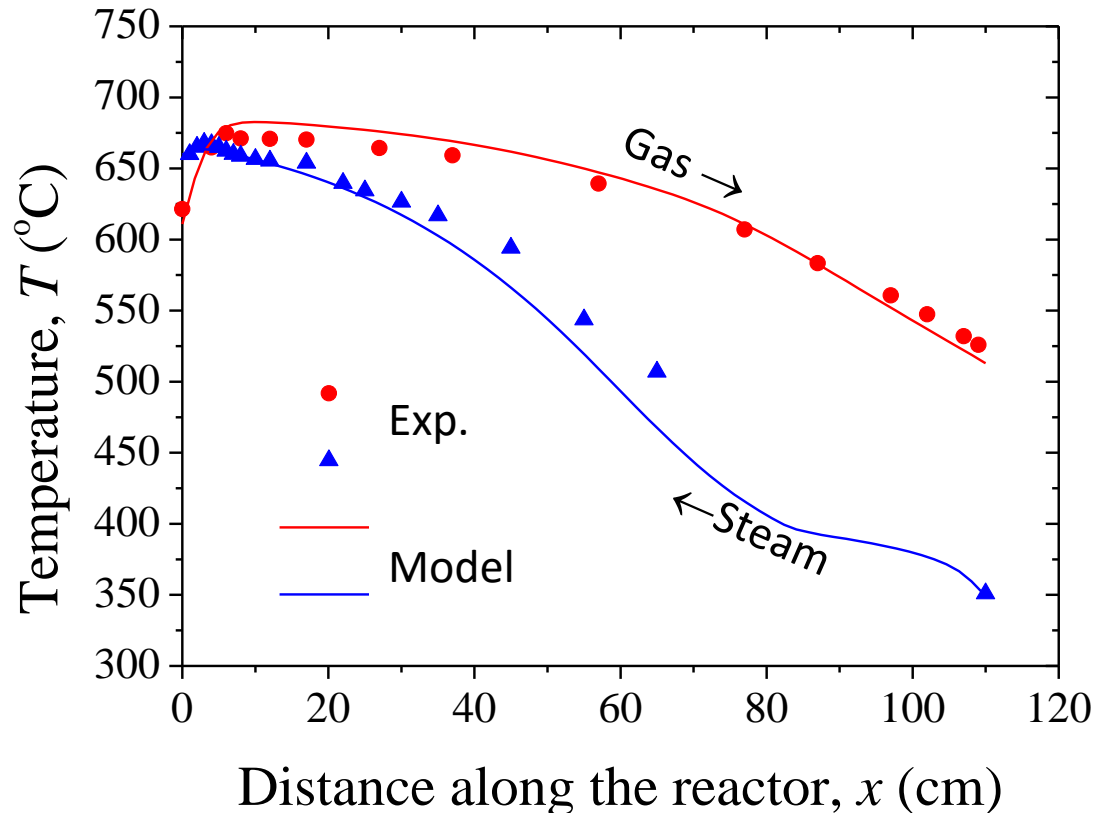
$$\dot{r}''' = \eta k_{0,m} \exp \left(-\frac{E_a}{R_u T} \right) \times \left[K_p^{-1} p_{N2} \left(\frac{p_{H2}^3}{p_{NH3}^2} \right)^\alpha p_o^{-(1+\alpha)} - \left(\frac{p_{NH3}^2}{p_{H2}^3} \right)^{1-\alpha} p_o^{(1-\alpha)} \right]$$

Supercritical Steam

$$(\dot{m} c_p)_s \frac{dT_s}{dx} = hP (T_{iw} - T_s)$$

$$Nu_b = C Re_b^m Pr^q \left(\frac{\rho_w}{\rho_b} \right)^{0.3} \left(\frac{\bar{c}_p}{c_{pb}} \right)^n$$

Success heating supercritical steam

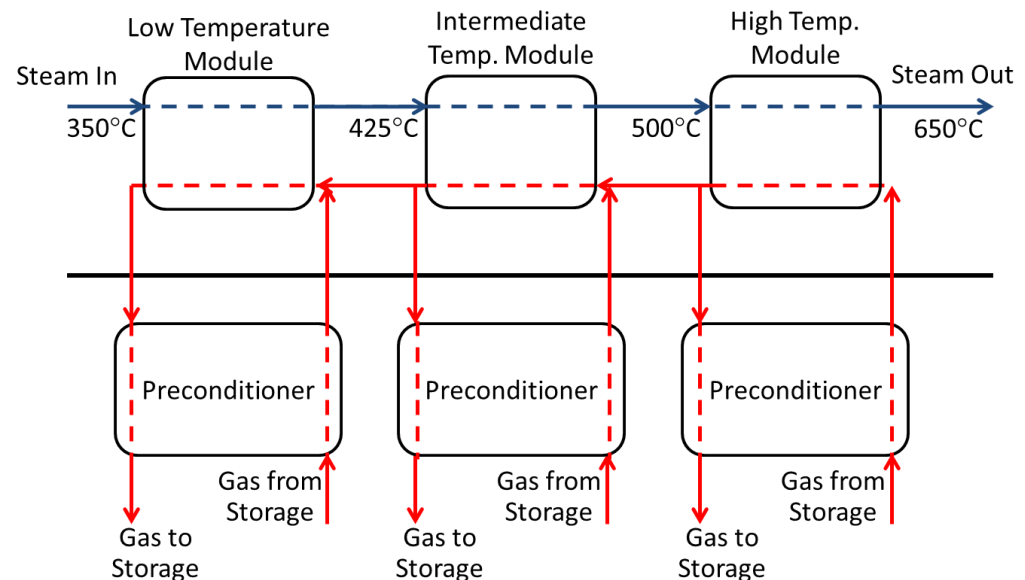


- Supercritical steam heated from 350 to 650°C.
- Model agrees well with experiment.
- Illustrated run: 1.6 kW_t
- Best run to date (with larger system): 4 kW_t

DESIGN OPTIMIZATION

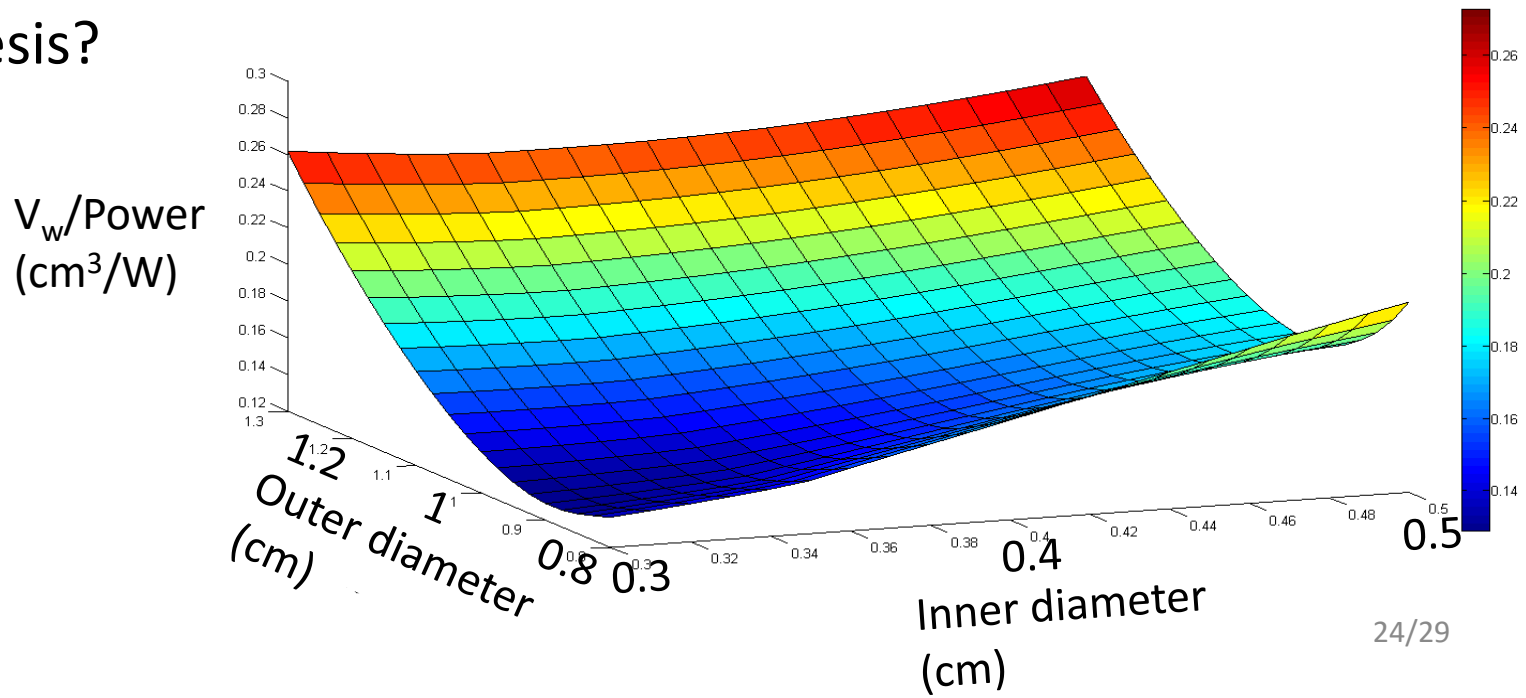
Multi-parameter optimization can improve design

- Consider entire synthesis system:
 - Synthesis reactor
 - Preconditioning subsystem
- Modular system with different reactor designs for different temperature regimes
- Multi-parameter optimization problem with tens of parameters.
- Largest cost is wall material.
- Minimize objective function:
 V_w/Power



Optimization drives to small scale

- Optimize inner and outer diameters – small is best.
- Not surprising – small scale improves heat transfer.
- How small can we go?
 - Pressure drop, pumping power considerations.
 - Manufacturing costs.
- Could micro-channel reactors be considered for ammonia synthesis?



Preliminary design: $V_w/\text{Power} \sim 0.7 \text{ cm}^3/\text{W}$

Multi-parameter optimization results:

- Single module, $V_w/\text{Power} = 0.09 \text{ cm}^3/\text{W}$
 - Smaller diameters (still commercially available)
 - Higher steam and gas flow rates per tube (but with reasonable ΔP constraints)
- Three module design, $V_w/\text{Power} = 0.05 \text{ cm}^3/\text{W}$
 - Smaller gas flow rates in high temperature/high reaction rate region
 - Higher gas flow rates in lower temperature/lower reaction rate regions

Full-scale system, 220 MW_t:

- **11 m³** wall material volume

Further optimization in progress.

- Identification of partners – current to next 12 months
- **Solar-driven** closed-loop experiment – 2017-2019
- Pilot 1 MW_e system – 2018-2021
 - **Gas storage** fabricated above ground using pressure pipe.
 - Heat recovery synthesis reactor designed for supercritical steam, but throttled for small off-the-shelf **subcritical steam turbine**.
- First utility scale demonstration, 10 MW_e – 2019-2024
 - First trial of **underground storage** using shaft drilling technology.
- First full-sized system, 100 MW_e, 10+ hrs storage – 2022-2027
 - Underground storage either salt cavern or drilled shaft.
 - **Potential for supercritical steam turbine.**

- Gas storage in salt caverns or drilled shafts appears feasible within the \$15/kWh_t budget.
- Ammonia synthesis can be used to heat supercritical steam to 650°C, according to experiments and modeling.
- Design optimization of the synthesis reactor system is underway:
 - Small diameter tubes are desirable.
 - Multi-parameter optimization of modular design has significantly decreased wall material volume.
- A proposed path-to-market could achieve a full-scale system by 2027.

- If ammonia is synthesized using intermittent energy sources (seasonal or diurnal), syngas storage may be needed.
 - Underground storage in salt caverns or drilled shafts might suit.
- Might the small-scale approach be useful for Haber-Bosch ammonia synthesis?
 - In the short-term before advanced technologies are developed?
 - To provide an approach that can scale down?
- Could high temperature heat recovery provide an advantage?
 - To supply heat for hydrogen production?
 - For efficient power generation?
- Is there merit to CSP-driven hybrid ammonia/electricity plant with storage?
 - 24-hr operation, optimizing time of electricity vs. ammonia production

Acknowledgments

- The information, data, and work presented herein was funded by the Office of Energy Efficiency and Renewable Energy, U.S. Dept. of Energy, Award No. DE-EE0006536. The authors gratefully acknowledge the support.

QUESTIONS?