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September 2016

Green Ammonia for energy storage and beyond

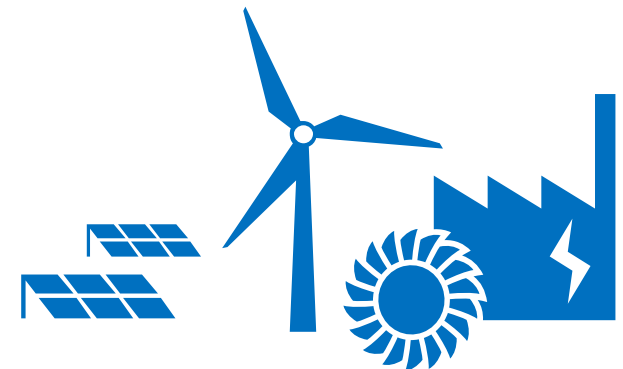
Ian Wilkinson

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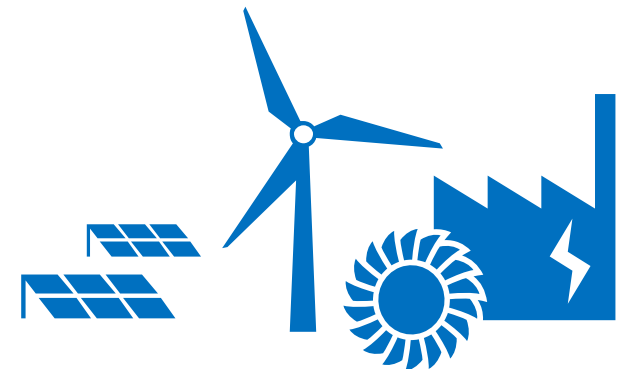
Green ammonia for energy storage and beyond

- Overview: Siemens perspective on energy storage and electrification
- Technology today:
Ammonia energy systems, and demonstrator project
- Technology developments:
Synthesis and energy release
- Prospects for “Green Ammonia”



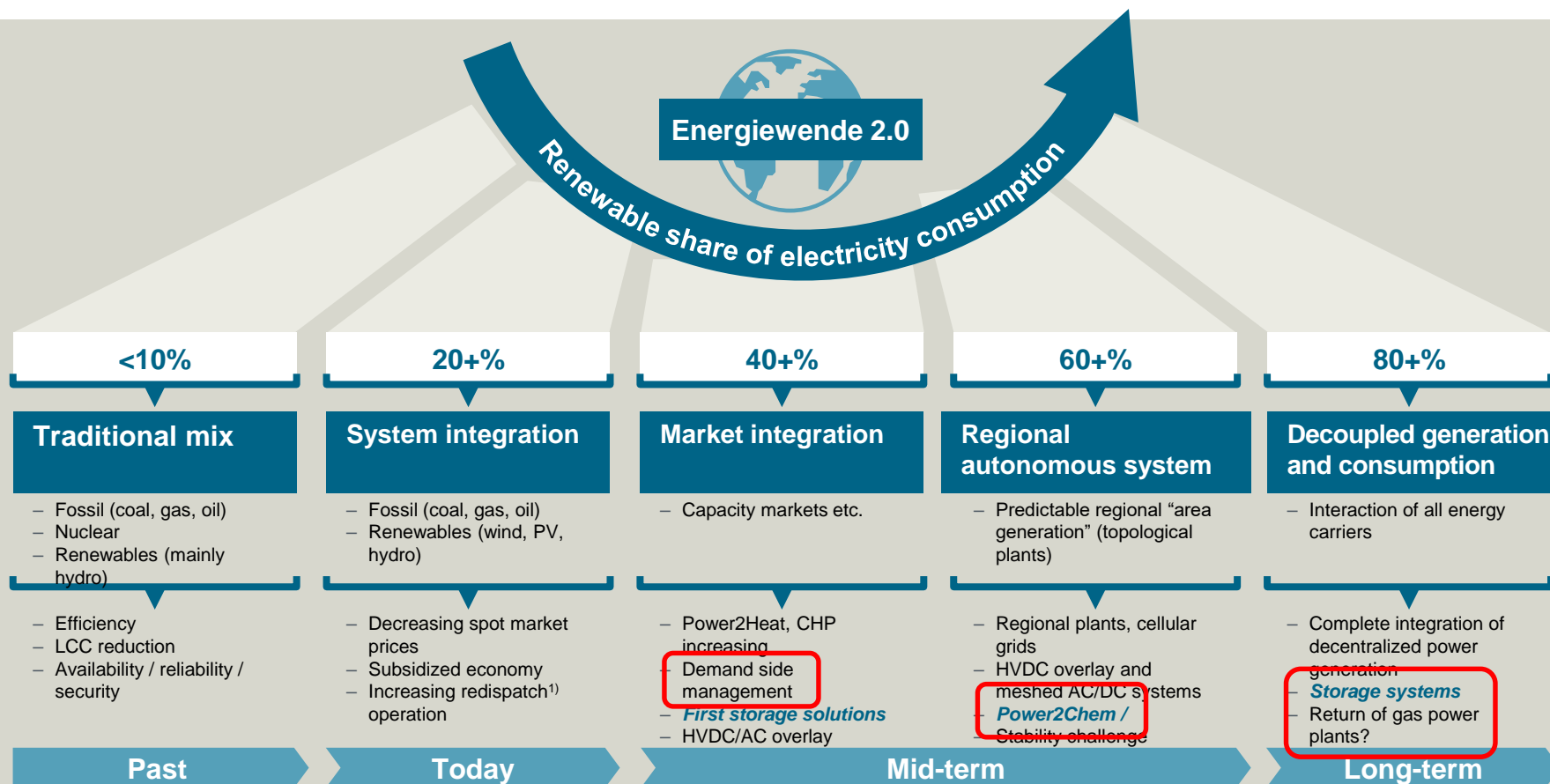
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The changing Energy Landscape

Different solutions for different market stages

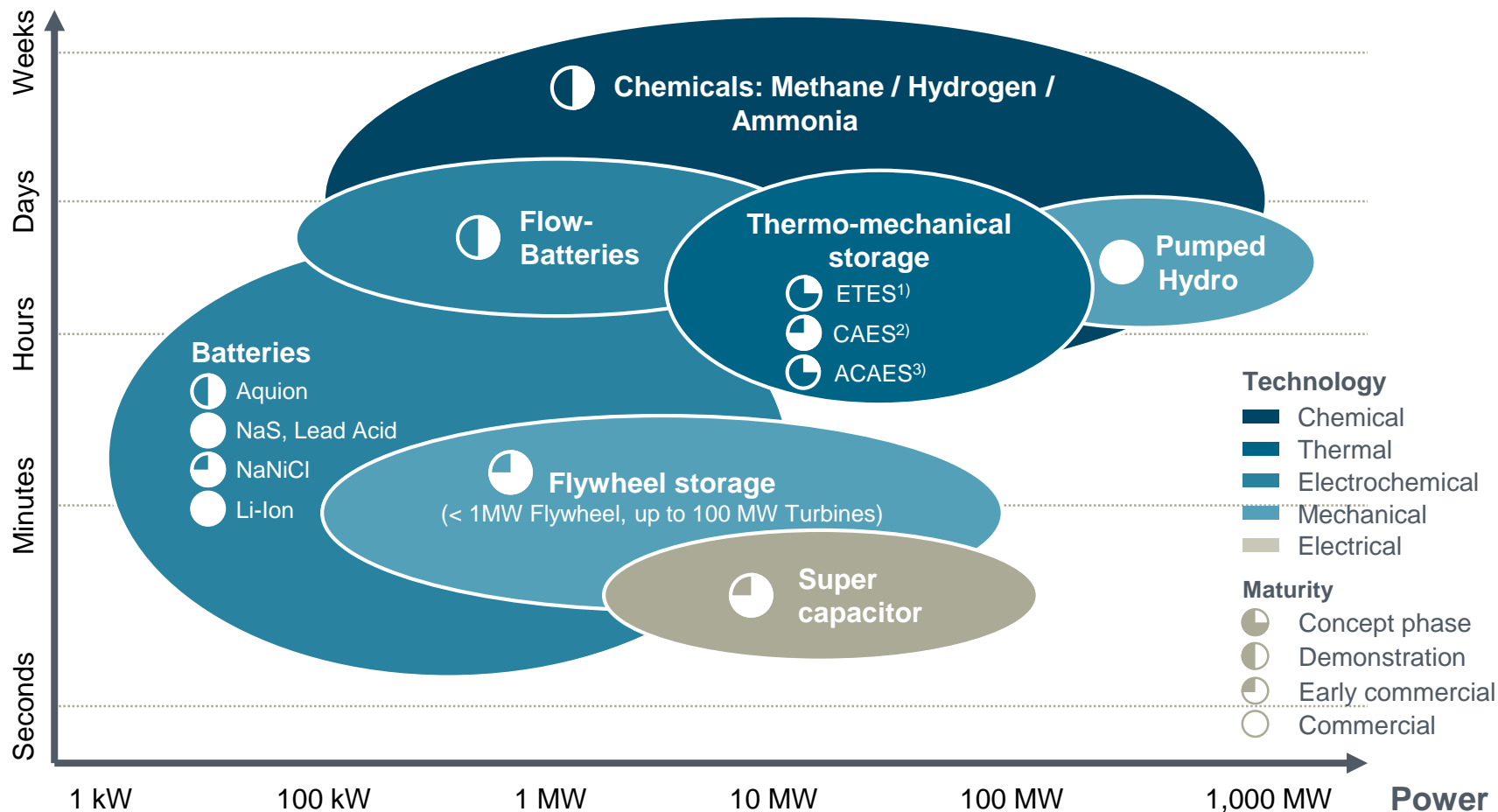


Fierce competition in traditional businesses, need to set benchmark in new or changed markets
Profitable business for new technologies cannot be shown yet – today’s use cases are mainly niche or pilot applications

1) Corrective action to avoid bottlenecks in power grid

Addressing energy storage needs will require a range of technologies

Storage time



¹⁾ Electro-Thermal Energy Storage

²⁾ Compressed Air Energy Storage

³⁾ Adiabatic Compressed Air Energy Storage

Energiepark Mainz – Project scope and key facts

- 3 x Sylizer 200 electrolyzers, with peak power of 2 MW el. each (6 MW peak).
- Connection to 10 MW wind-farm
- 1000 kg H₂ storage (33 MWh)
- Provides:
 - H₂ gas to Linde
 - Demand side response to local grid
 - H₂ gas injection into local (natural) gas network

**ENERGIE
PARK MAINZ**



The existing chemical industry emissions conflict with initiatives to avoid climate change

Chemical Industry Emissions

1255 MT/yr CO₂¹
→ 4% world total²

1.1TW¹
→ 8.2% world total²

≠

Climate Act Requirements

UK target of 80% cut in emissions by 2050

EU wide target of 40% cut in emissions by 2030

Opportunity: carbon – free synthesis of chemicals powered by renewable energy

Ammonia: 1.8% of the world consumption of fossil energy goes into the production of ammonia. 90% of ammonia production is based on natural gas.

Top 10 Chemicals / Processes:

1) Steam cracking

2) Ammonia

3) Aromatics extraction

4) Methanol

5) Butylene

6) Propylene FCC

7) Ethanol

8) Butadiene (C4 sep.)

9) Soda ash

10) Carbon black

1) Chemical and Petrochemical Sector – IEA2009 2) Key World Energy Statistics – IEA2014

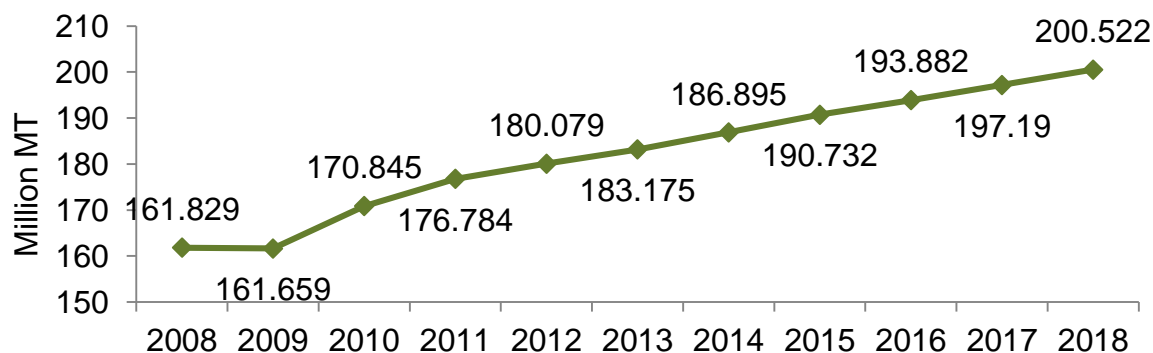
Ammonia is an important chemical with a commodity market value of EUR100bn/year

Ammonia

- A gas, produced by the chemical industry. Over 80% of ammonia is used in the fertiliser industry.
- Demand for fertiliser, as shown in the graph (including projected growth to 2018), is growing at +3%pa¹.
- Current production levels of Ammonia are about 180m t/year. The commodity value is €600-€700/t, leading to a commodity market value of over €100bn/year
- Production today uses the Haber-Bosch process and relies on natural gas as a feedstock.

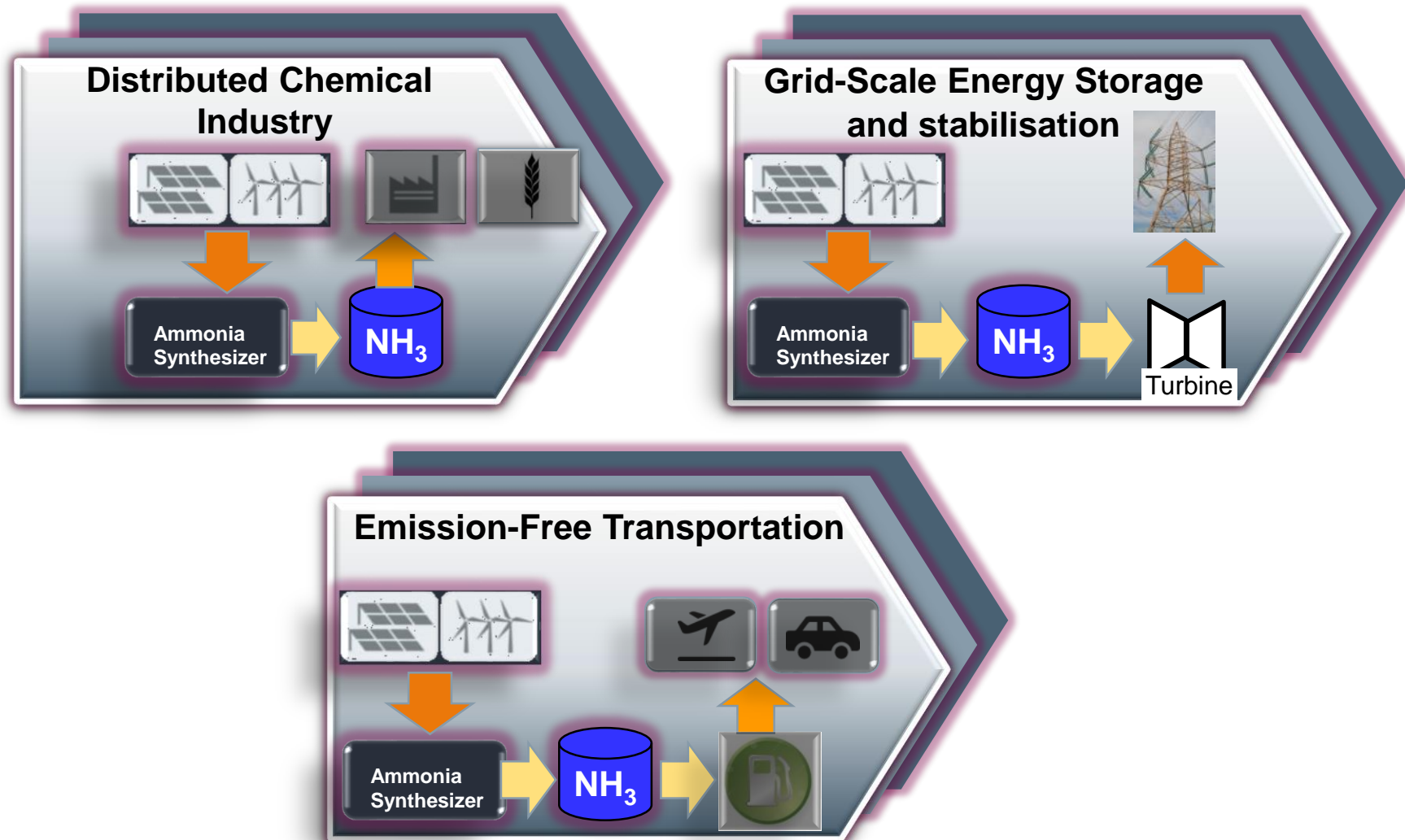


Global fertilizer nutrient consumption



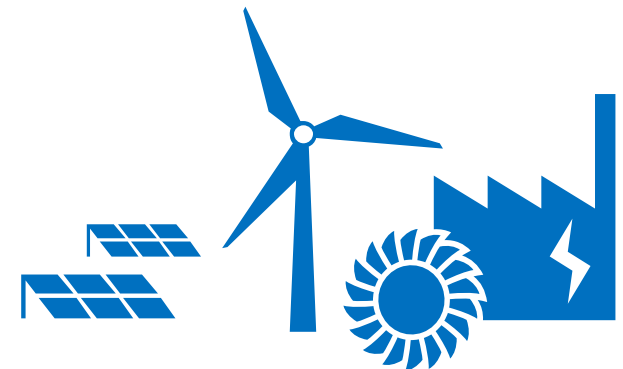
Source: World Fertilizer Trends and Outlook to 2018, Food and Agriculture Organization of the United Nations

Green Ammonia – Carbon Free Flexible Asset



Green ammonia for energy storage and beyond

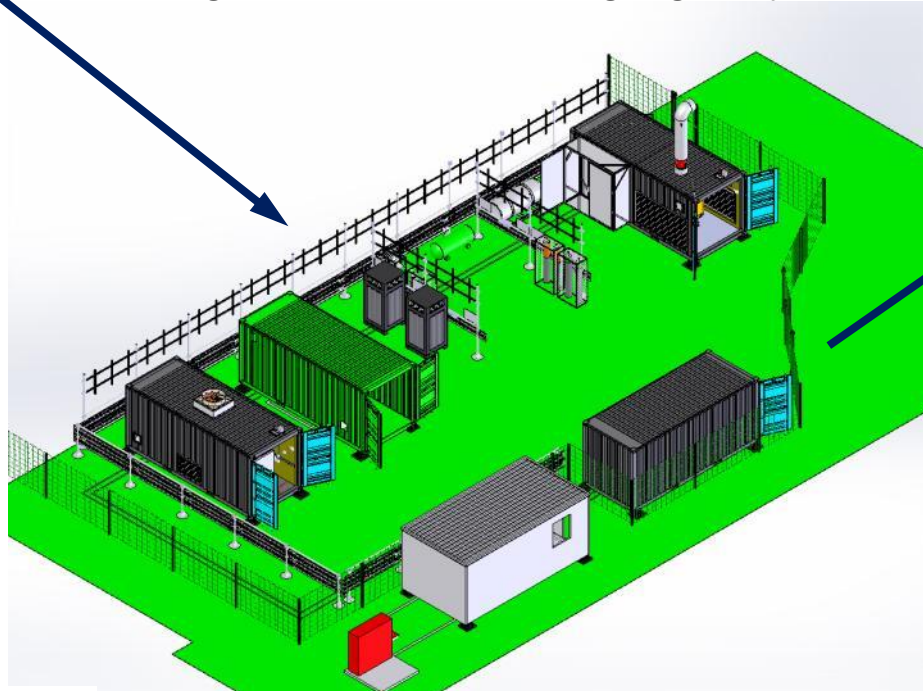
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Decoupling Green Energy: “green” ammonia synthesis and energy storage system demonstrator

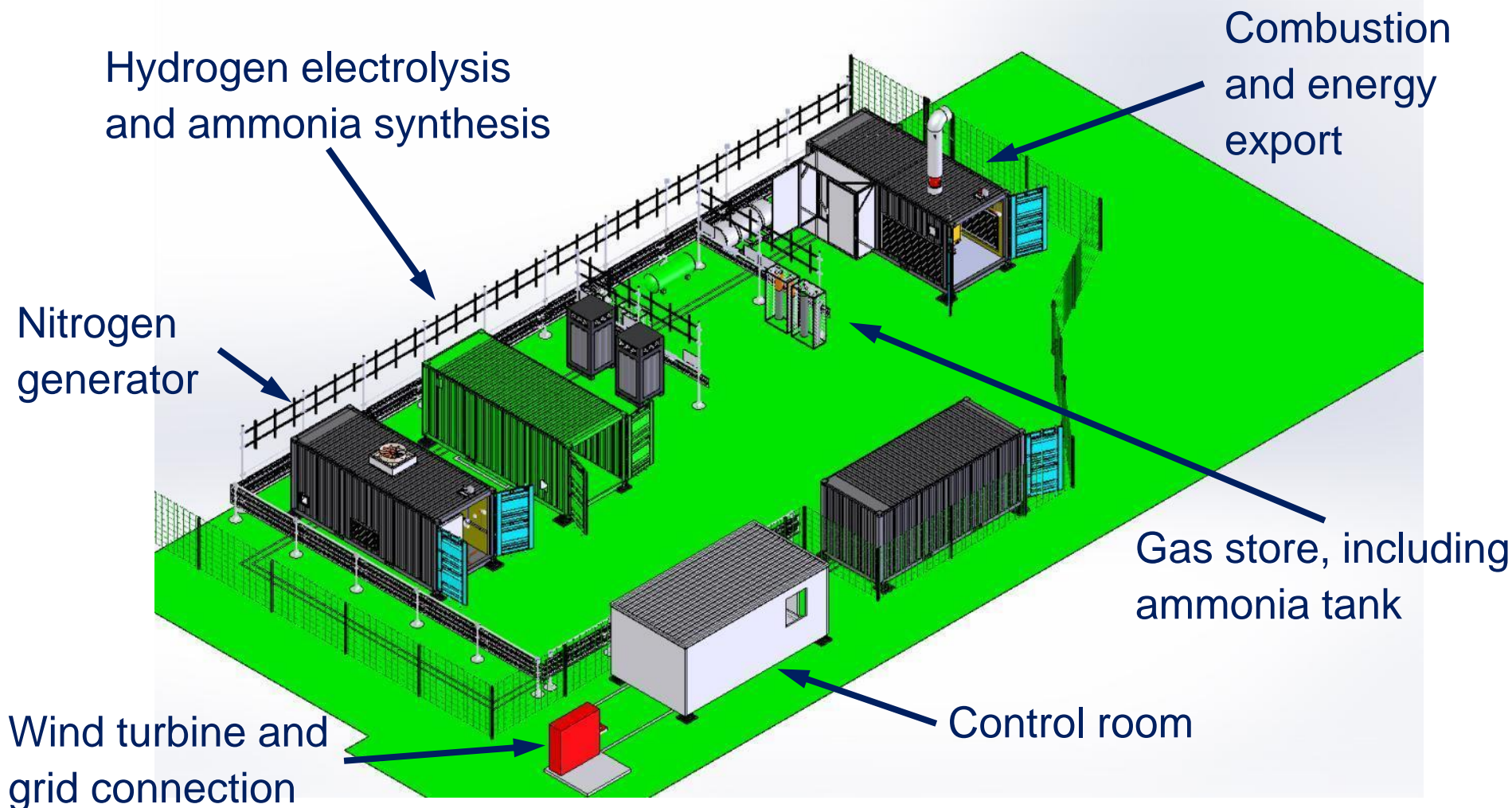


- Being built at Rutherford Appleton Laboratory, near Oxford, UK.
- Project 50% supported by Innovate UK (UK government funding agency).



- Evaluation of all-electric synthesis and energy storage demonstration system by Dec 2017.

Ammonia test site layout



Ammonia energy storage system demonstrator hardware



SUBJECT:
MODEL TYPE

Nitrogen On-Site generator Plant
LASERCUT QN2 225-1

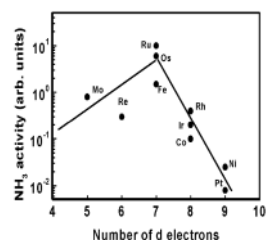
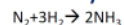


**NH₃ H-B
Reactor
(Bespoke)**

System demonstrator technology development

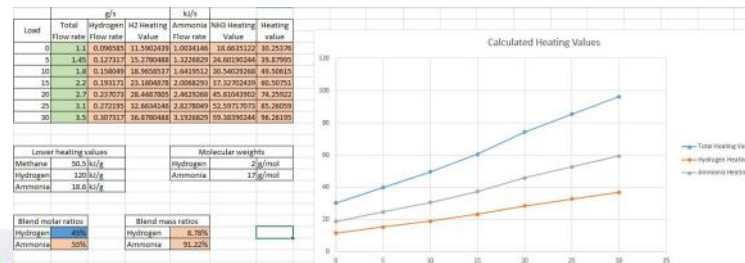
Haber-Bosch synthesis catalyst

Ammonia synthesis



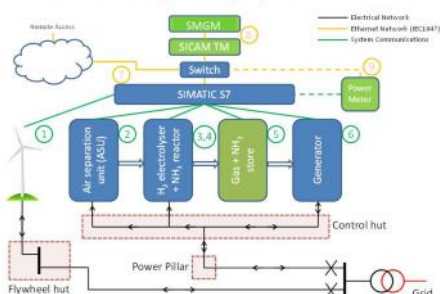
Ozaki and Aika, Catalysis 1 (Anderson and Boudart, Ed.)

Ammonia combustion studies

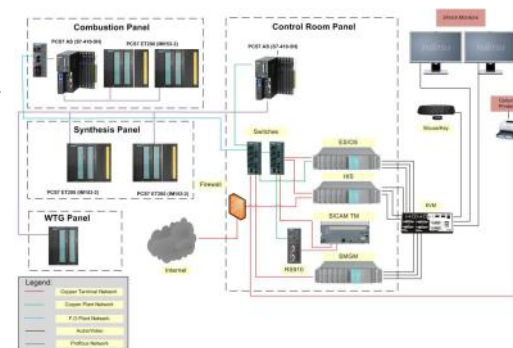


Energy management system

Ammonia energy-storage system overview

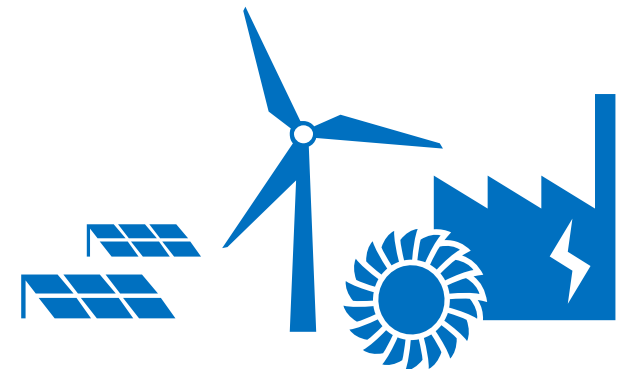


Control system



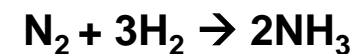
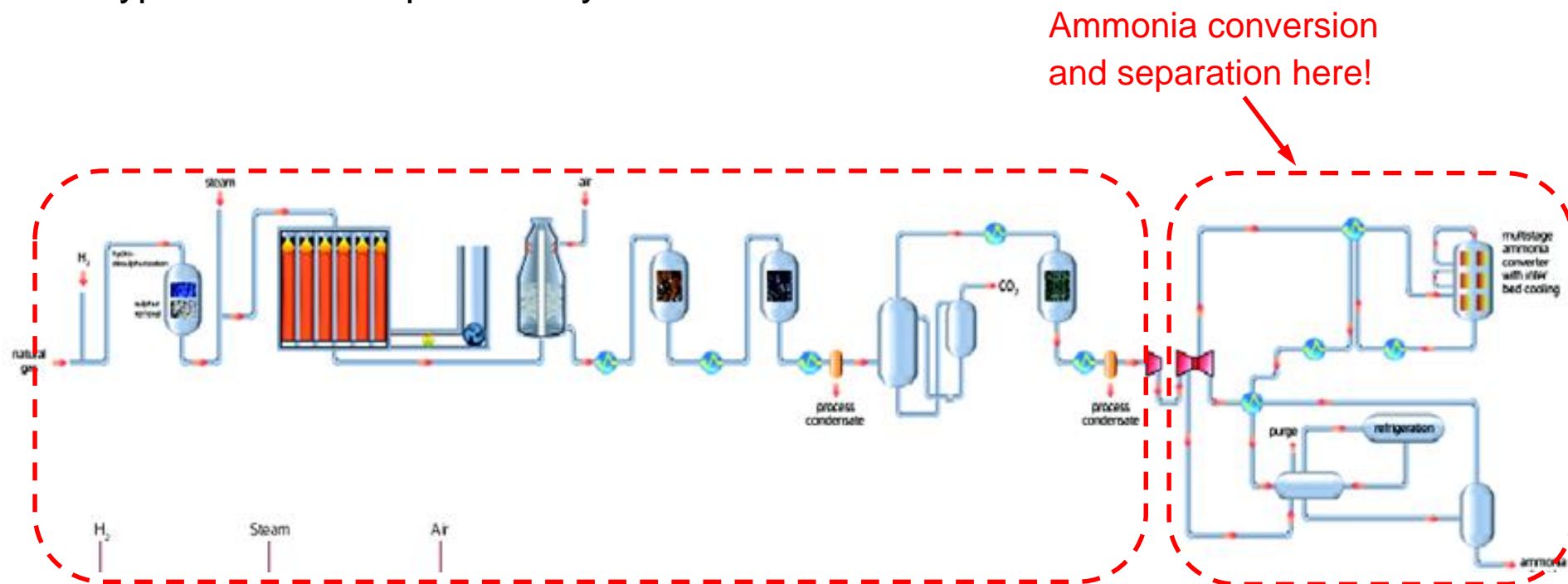
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Ammonia Production Today

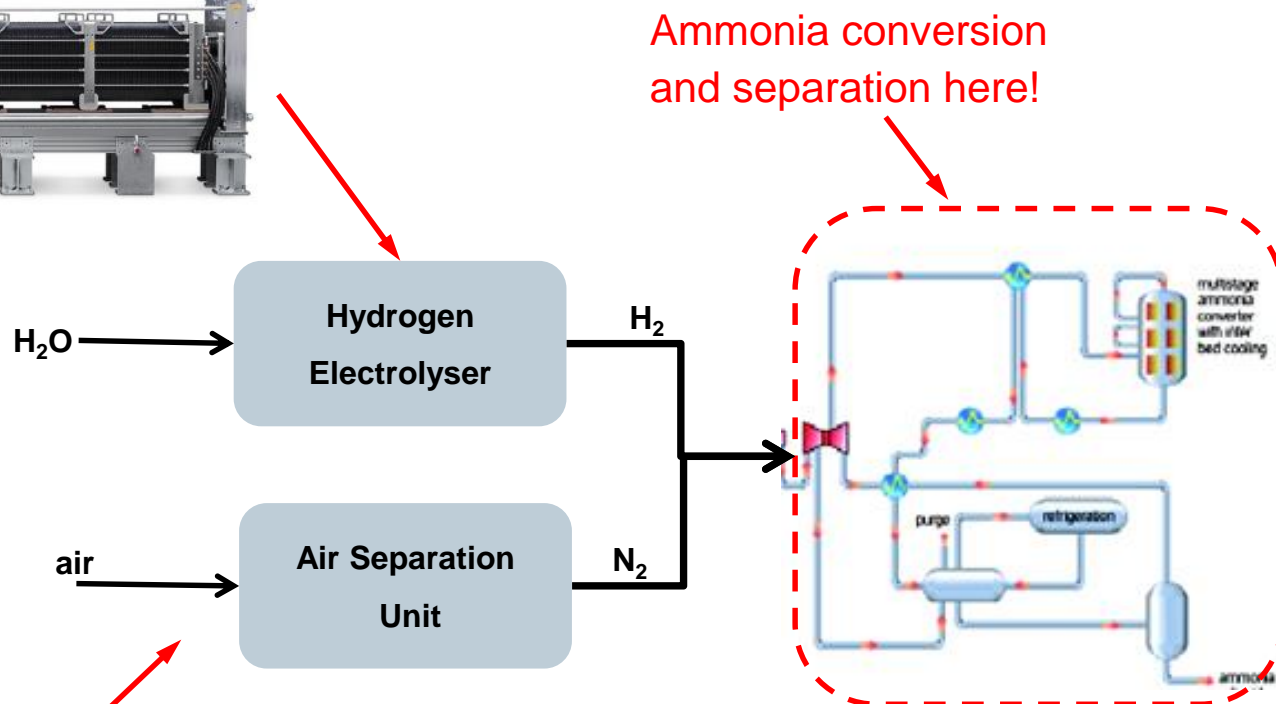
Typical ammonia plant today¹



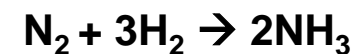
Ammonia Production 2020



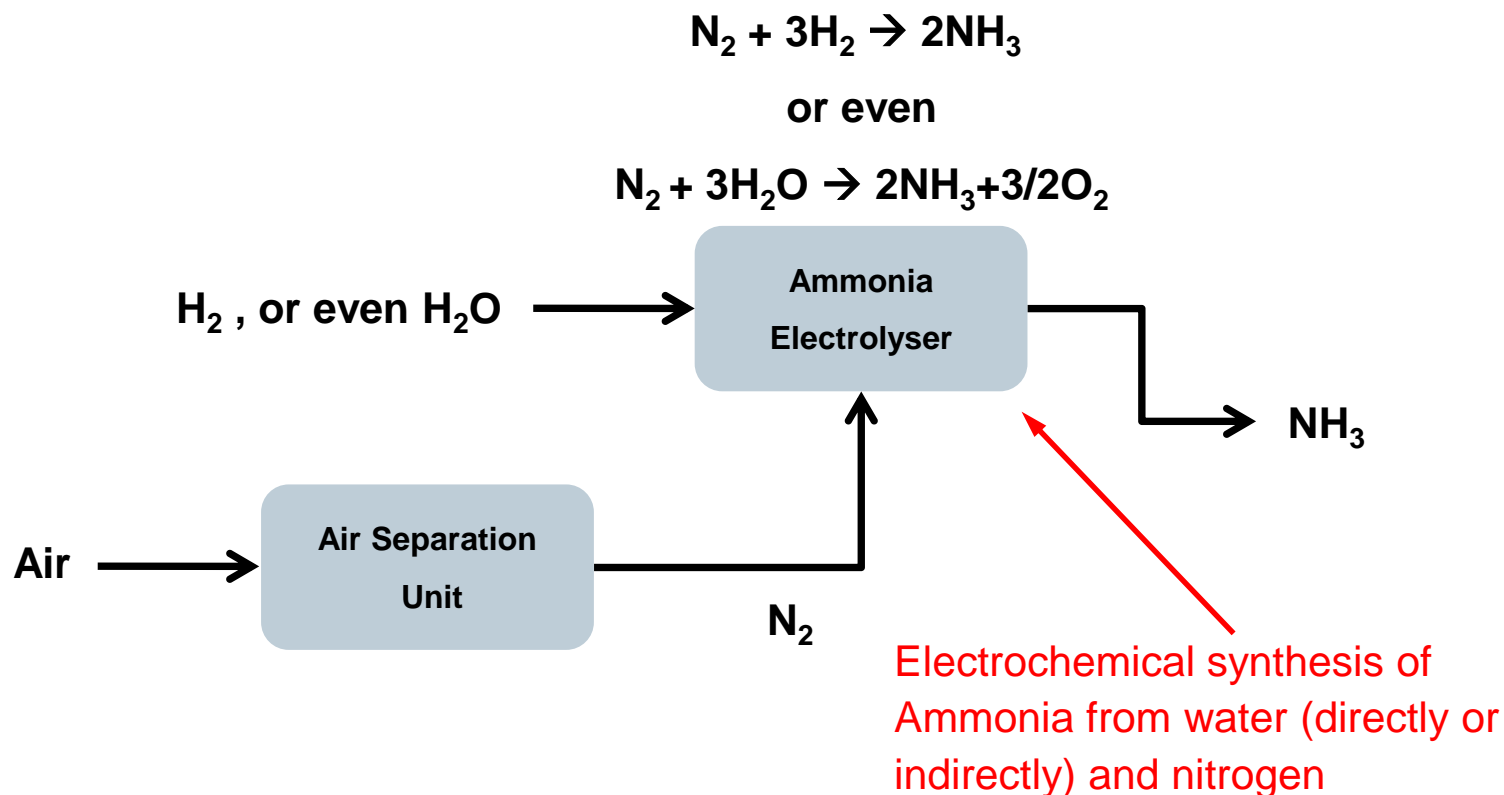
Siemens Sylizer



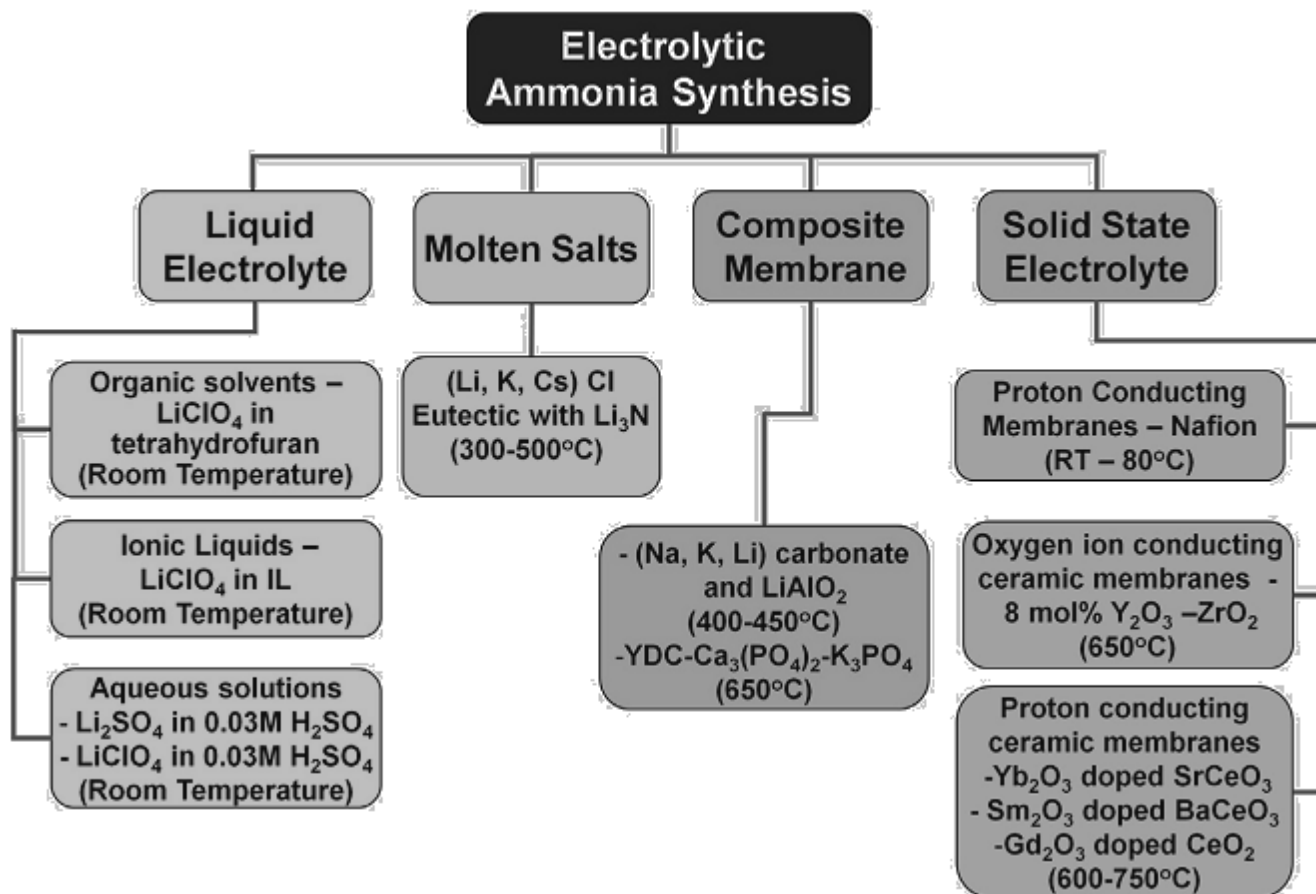
Gas preparation: ultra pure feedstock gas from water electrolysis and air separation unit



Ammonia Production 2030



Electro-catalytic synthesis of NH₃ – 4 focus areas



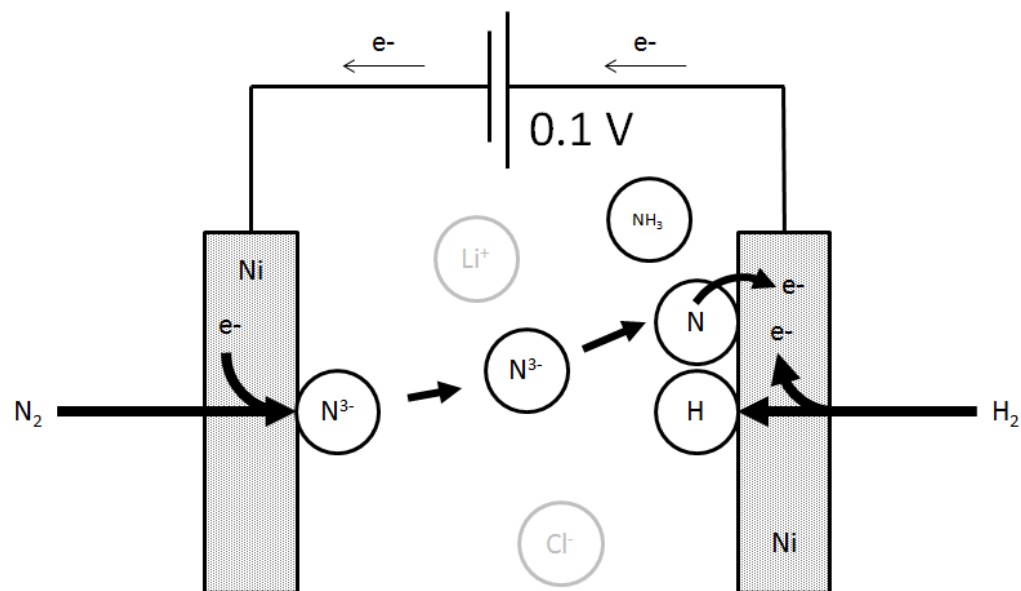
Molten Salt Approach

Stability of N^{3-} in metal halide salts allows direct reduction of N_2 to N^{3-} at ambient pressure

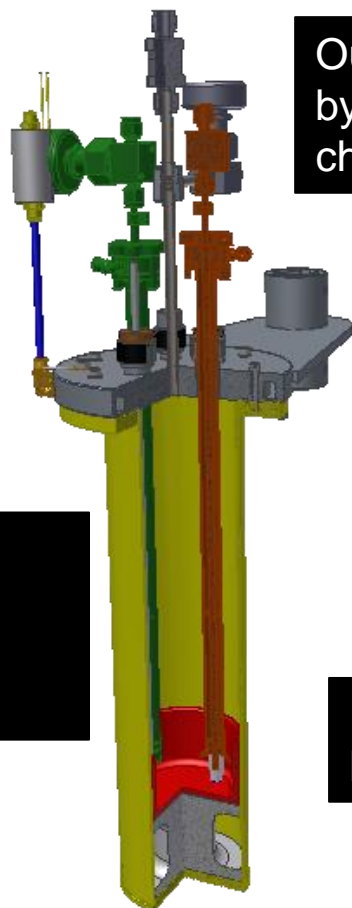
Applied voltage causes migration of N^{3-} from surface of negative electrode to surface of positive electrode

Facile dissociation of H_2 occurs on positive electrode to generate surface H
Surface N and H combine to produce ammonia

Equivalent to high pressures used in thermal route



Molten Salt Experimental Program



Gas supplied to porous electrodes (orange and green tubes)

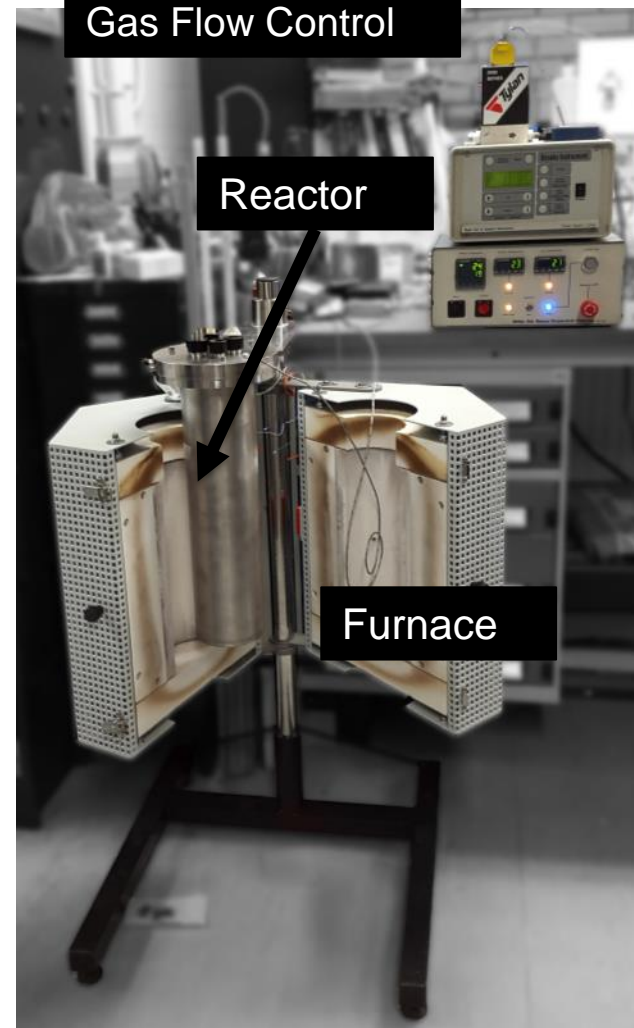
Outlet gas analysed by gas chromatography

100 mL molten salt held in crucible

Temperature and Gas Flow Control

Reactor

Furnace



Solid Electrolyte Ammonia Electrolysis

Anode

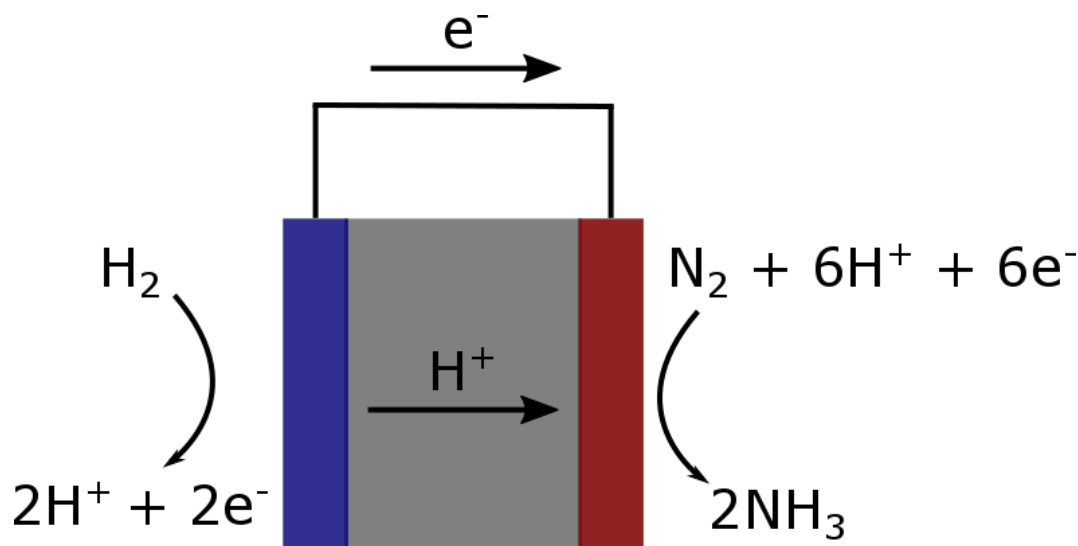
- Oxidation of hydrogen
- Electrically conducting
- Proton conducting

Electrolyte

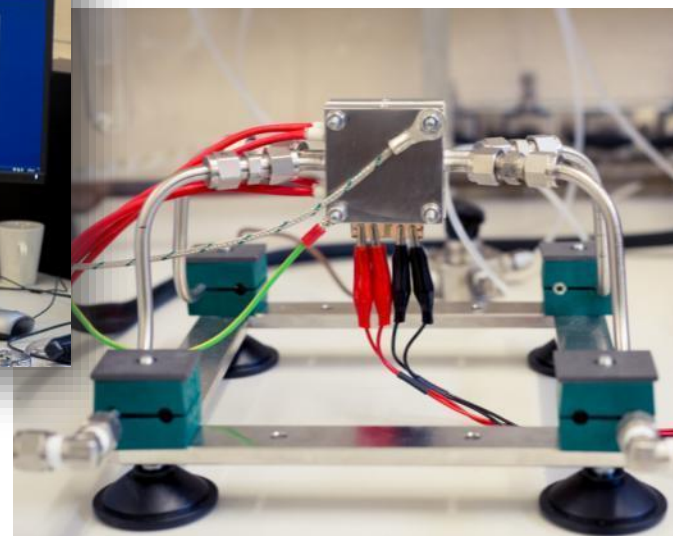
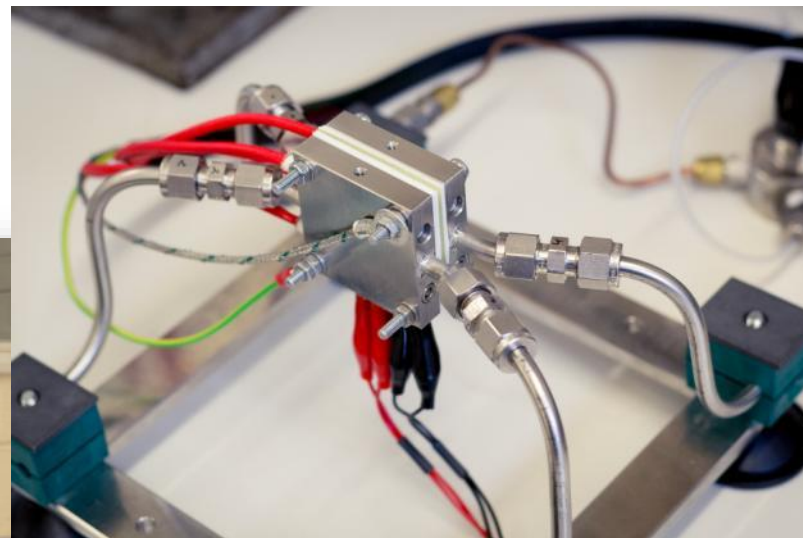
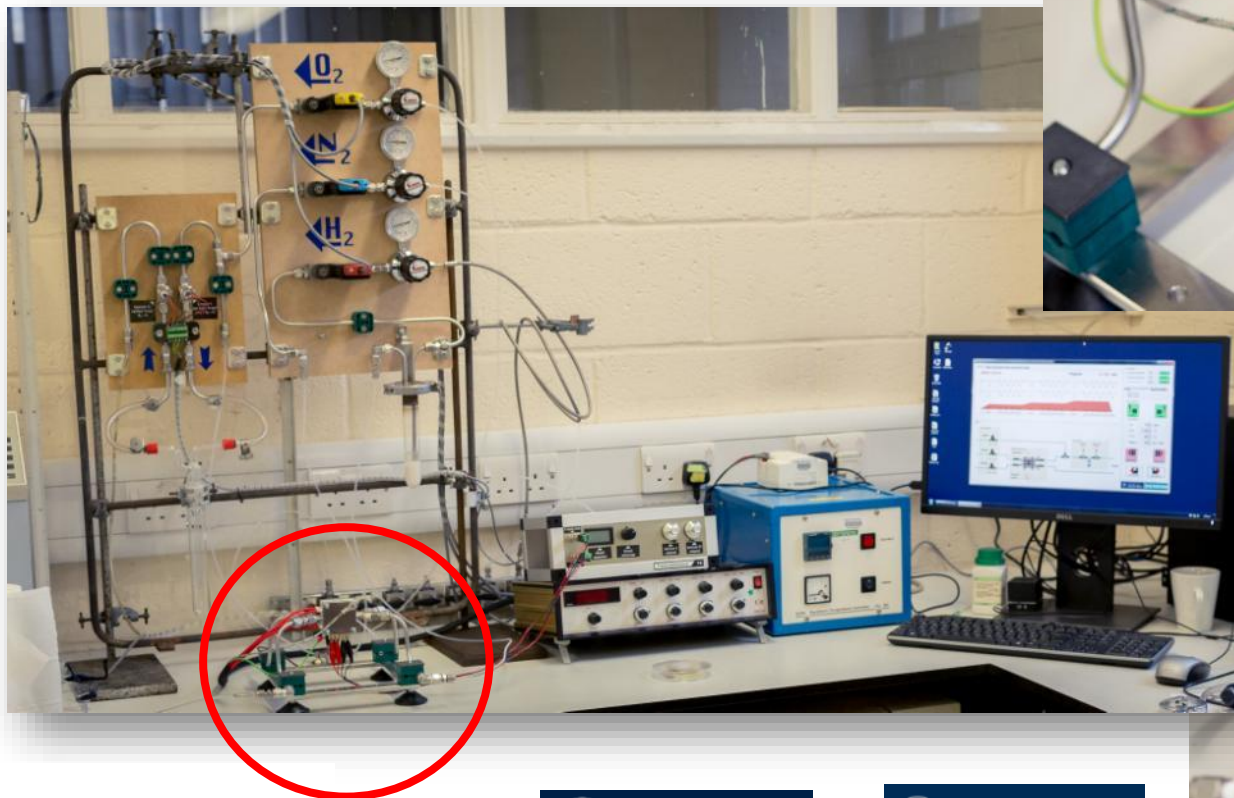
- Proton conducting
- Electrically insulating

Cathode

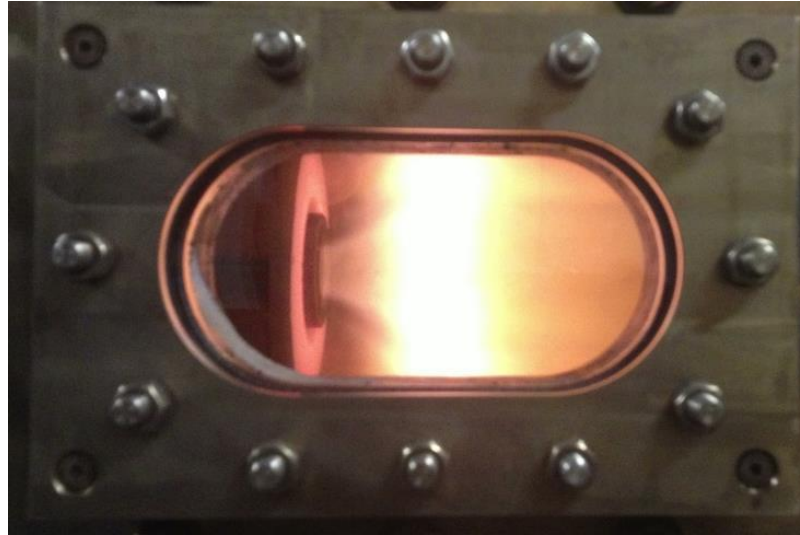
- Reduction of nitrogen
- Formation of ammonia
- Electrically conducting
- Proton conducting



Solid electrolyte experimental programme



Preliminary studies of ammonia combustion for gas turbines

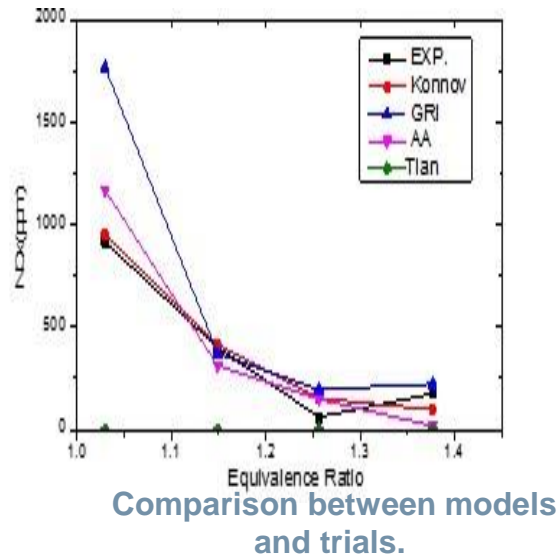


- Flame with a 62% ammonia, 38% methane blend at an air / fuel equivalence ratio of 1.31 in Cardiff's HPOC test rig.
- High equivalence ratios showed a reduction in NO_x emissions, but suffered unstable flames in high swirl regime of conventional industrial combustor.

Ammonia Gas Turbine Development



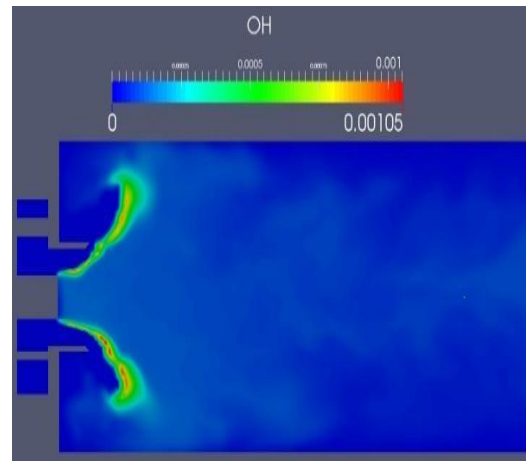
Generic burner, high pressure.



Comparison between models and trials.



Lab combustor. Thermoacoustics.

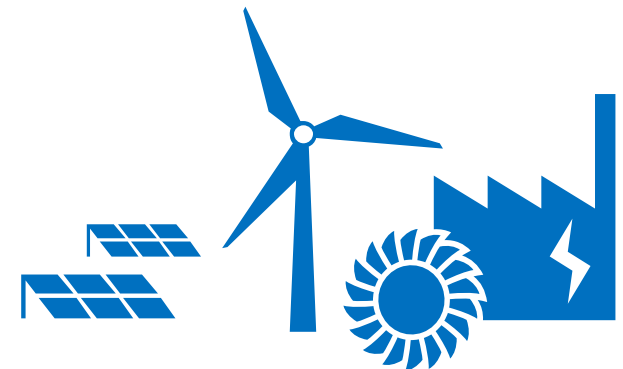


CFD model using $\text{NH}_3\text{-CH}_4$ with GRI-Mech

- Evaluation of current reaction models to determine accuracy and restrictions.
- Modelling of generic swirl burners through CFD studies to study combustion and emission patterns.
- Recommendation of first ideas for technology improvement: stratified injection.

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Development path for ammonia synthesis and energy release

2015

Centralised Haber-Bosch synthesis



NH3 internal combustion engine

2020

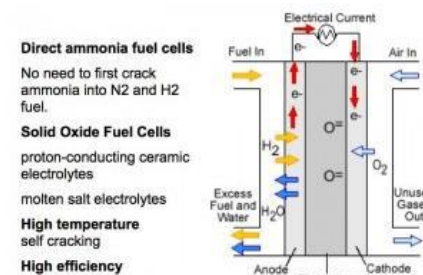
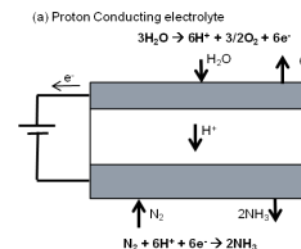
All electric Haber Bosch dynamic, distributed



NH3 Gas Turbines

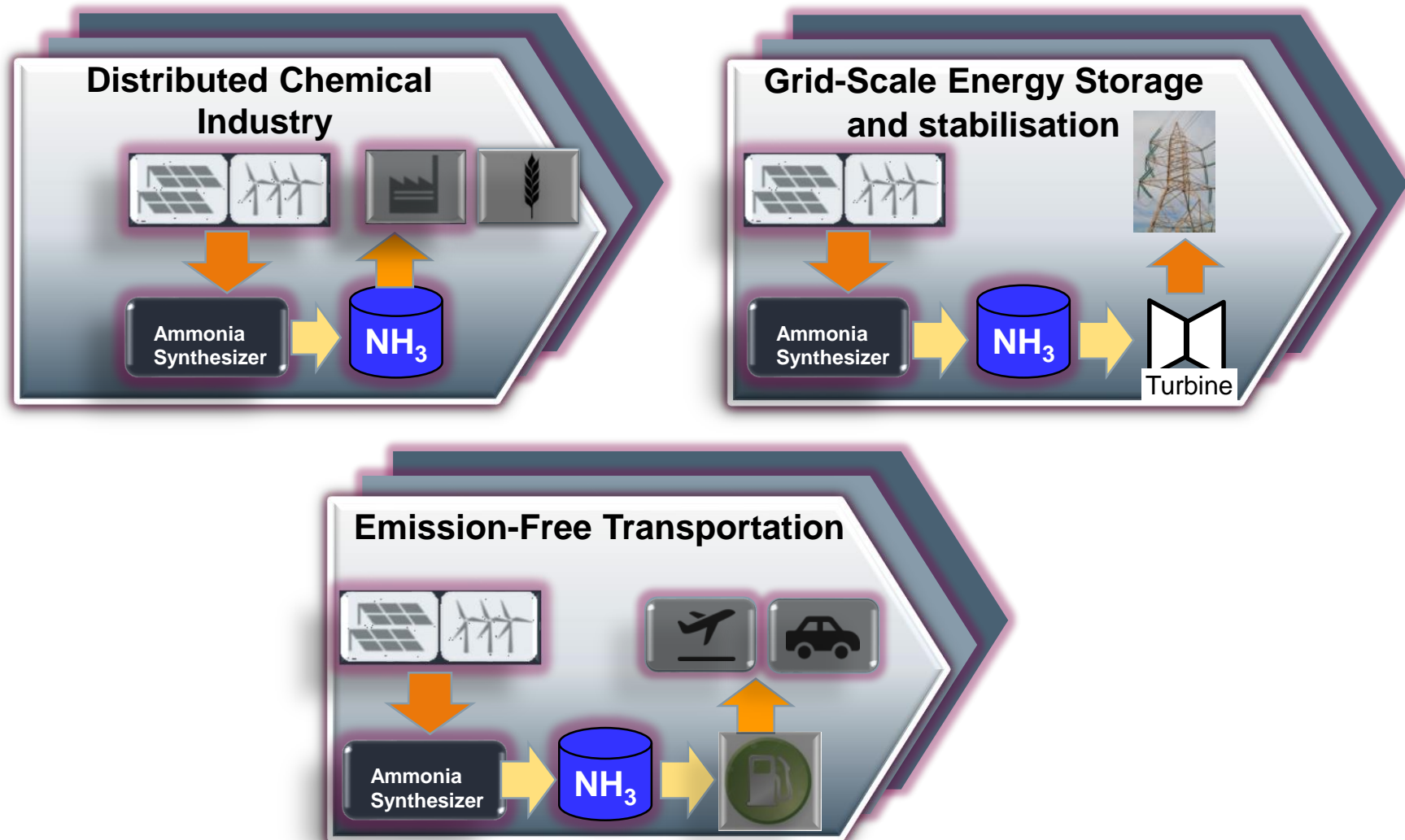
2025

Large scale, distributed electrochemical synthesis



NH3 Fuel Cell

Green Ammonia – Carbon Free Flexible Asset



For further information, see

<http://www.siemens.co.uk/green-ammonia>

'Green' Ammonia

Green ammonia is key to meeting the twin challenges of the 21st century.

Ammonia

Ammonia is a compound made of nitrogen and hydrogen. Chemical formula NH_3 . Ammonia's main use is in fertilizer.



Nitrogen is a harmless odourless gas that makes up 78% of the air around us.

Hydrogen is the most abundant element in the universe. There are 2 hydrogen atoms in every molecule of water.



By using water electrolysis and renewable electricity, ammonia production can be made completely carbon-free.



By 2050 there will be ten billion people on the planet.



Using ammonia as fertilizer makes land more productive. Increasingly vital as the population grows and living standards improve.

People need food and energy and it must be CO_2 free – that's where green ammonia comes in.

Ammonia



Ammonia feeds the world: 180 million tonnes were produced in 2015, mainly for use in fertilizers. Growing demand for food means this must rise 3% each year.



Today, ammonia is made using the Haber-Bosch process invented and perfected in the early 1900s. Its two inventors won Nobel prizes in 1918 and 1931.

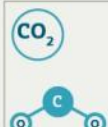


In the Haber-Bosch process hydrogen and nitrogen are converted to ammonia using high temperature and a catalyst.



The global trade in ammonia means we already know how to transport and store it safely.

But there is a problem



Today the lowest cost way to get hydrogen is from natural gas but this produces carbon-dioxide (CO_2), which is a cause of man-made climate change.



Ammonia production requires energy, and today this energy also comes from fossil fuels. Together with the fossil hydrogen feedstock, current ammonia production accounts for over 1% of global CO_2 emissions.



To keep under 2 degrees warming humans must emit no more than 600 billion tonnes more CO_2 . That's less than 25 years at the rate today.



Once CO_2 is released into the atmosphere it will change the climate for the next 10,000 years.

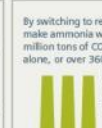
The good news – innovating to create carbon free "green" ammonia



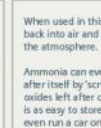
We can make hydrogen from water using electricity – a process known as electrolysis. Passing an electric current through water separates the hydrogen and oxygen.



Green ammonia can support the business case for renewables, by providing an alternative revenue stream that is not dependent on a grid connection (particularly relevant in remote areas), and by being used for load-balancing.



By switching to renewable electricity to make ammonia we could save over 40 million tons of CO_2 each year in Europe alone, or over 360 million tons worldwide. We can also burn ammonia to make electricity when the wind is not blowing.



When used in this way ammonia turns back into air and water, so doesn't pollute the atmosphere.

Ammonia can even be made to clean up after itself by 'scrubbing' any nitrogen oxides left after combustion. Ammonia is as easy to store safely as LPG. You can even run a car on ammonia.

Contacts



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CT REE

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