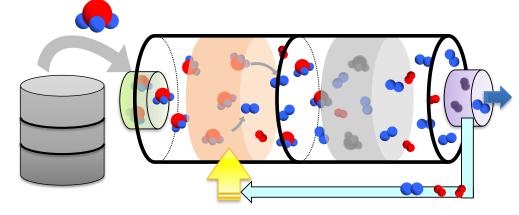
# Ammonia As a Hydrogen Carrier for PEM Fuel Cells



18 AIChE Annual Meeting in Pittsburgh, PA, October 28-November 2, 2018, Wednesday, October, 31, 2018 f rom 9:45 - 10:00 David L. Lawrence Convention Center, 317

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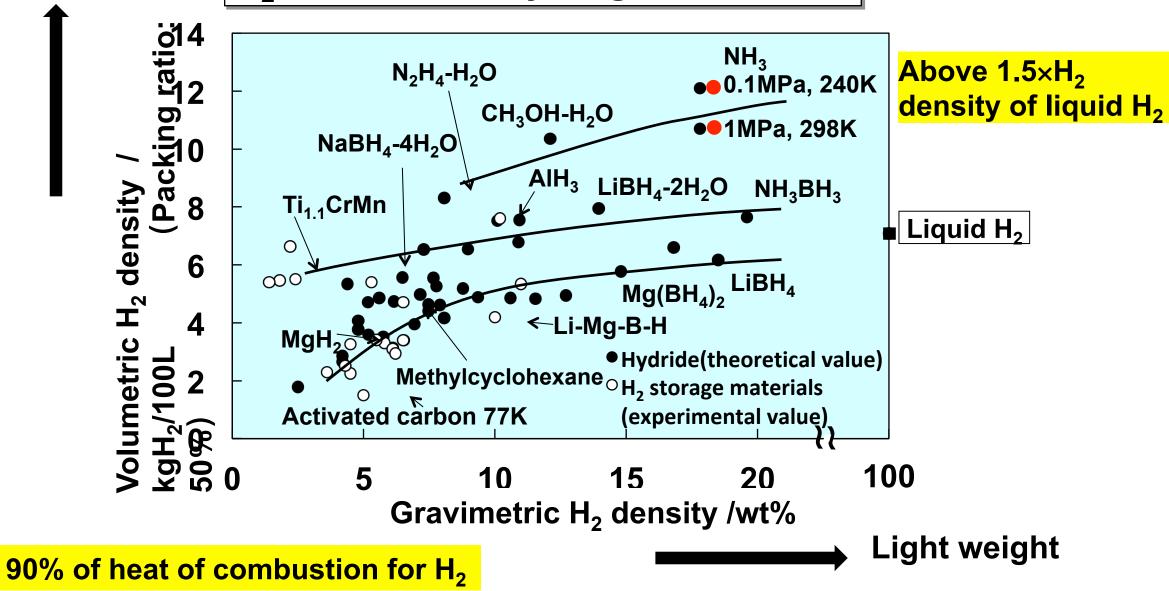






Compact

H<sub>2</sub> densities of hydrogen carriers



# 2. Basic technology for hydrogen station utilizing ammonia

**Excellent storage capacity and high heat of combustion** Hydrogen station utilizing ammonia: Key issue

Specification of hydrogen fuel for FCV (ISO 14687-2:2012)

| Species                  | Concentration |
|--------------------------|---------------|
| Purity of H <sub>2</sub> | 99.97%        |
| N <sub>2</sub> , Ar      | 100ppm        |
| Ammonia                  | 0.1ppm        |

History of this research (press release, July, 19 2016)

Component technologies to produce high-purity hydrogen from ammonia (NH<sub>3</sub> <0.1ppm, N<sub>2</sub> <1ppm, H<sub>2</sub> >99.98%)

(1)NH<sub>3</sub> decomposition catalyst (Ru/MgO, Ru: 3wt%) and 1Nm<sup>3</sup>/h-scale simulated shell and tube cracker: Low heat transfer performance

(2)NH<sub>3</sub> storage material (zeolite) and 1Nm<sup>3</sup>/h-scale remover: NH<sub>3</sub> <0.1ppm

(3) 1Nm<sup>3</sup>/h-scale H<sub>2</sub> purifier (two-tower-type): Purification efficiency of 70%

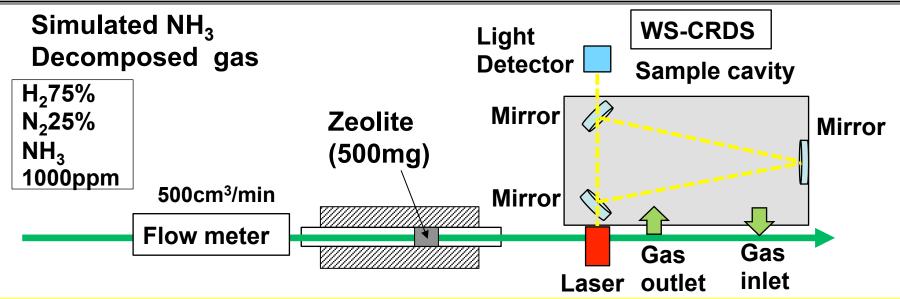
# **Purpose of this research**

- Component technologies to detect ammonia concentration smaller than 0.1 ppm
- 2. Component technologies to improve purification efficiency and H<sub>2</sub> conversion efficiency without emission of CO<sub>2</sub>

# (1) NH<sub>3</sub> storage materials and remover

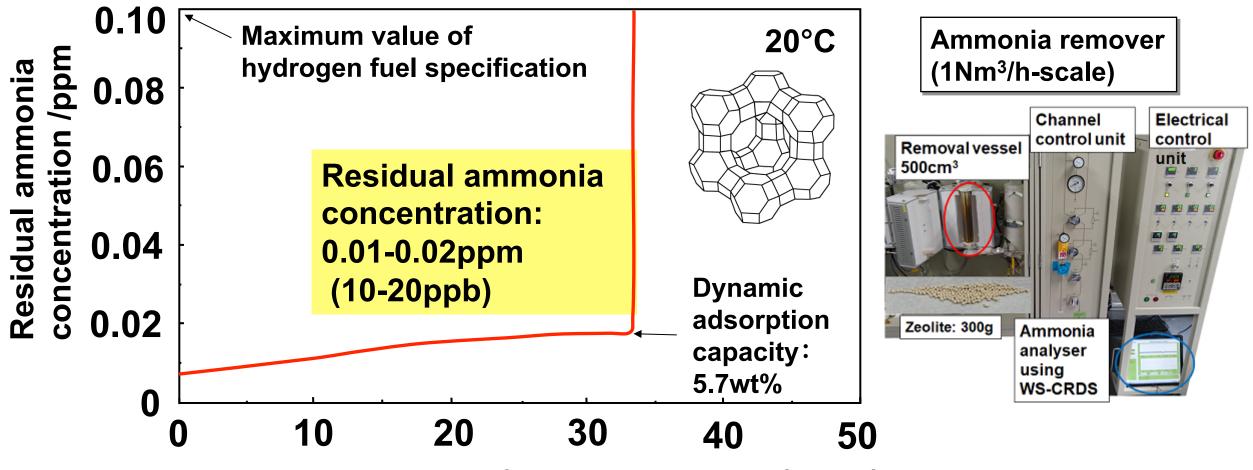
Zeolite: High adsorption power, flow channel, recyclable by heating

Conceptive picture of specially designed breakthrough testing apparatus



Exit NH<sub>3</sub> concentration: Cavity ring-down spectroscopy is a reliable technique to measure trace ammonia in gases even at ppb level.

### Residual ammonia concentration passed through zeolite packed column

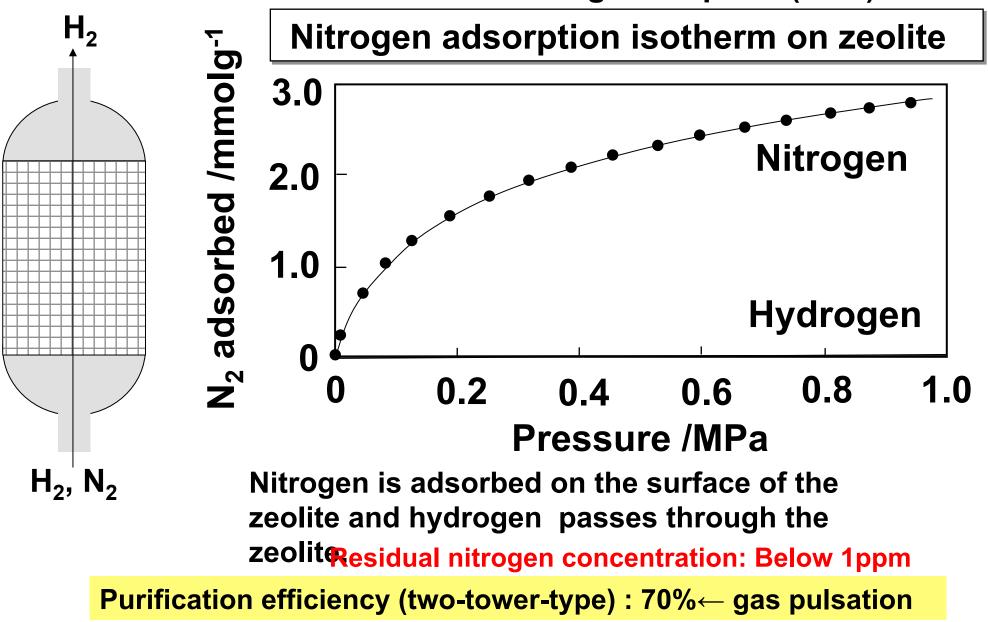


Accumulation amount of ammonia supply /gNH<sub>3</sub>/L

NH<sub>3</sub> concentration: Sufficient satisfaction of hydrogen fuel specification

#### (2) H<sub>2</sub> purifier with off-gas supply unit





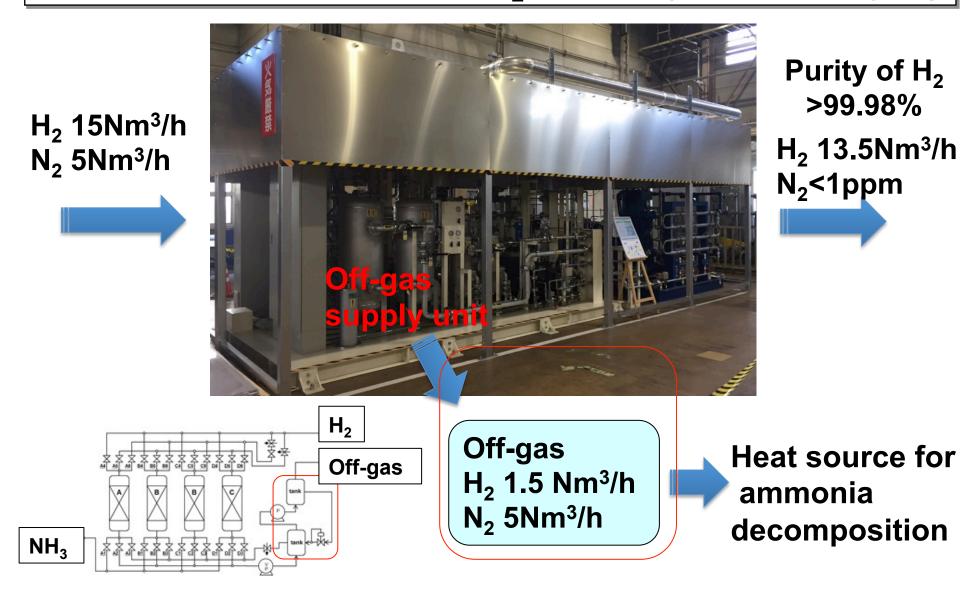
# Overview of 1Nm<sup>3</sup>/h-scale H<sub>2</sub> purifier (four-tower-type)



#### **Off-gas supply unit**

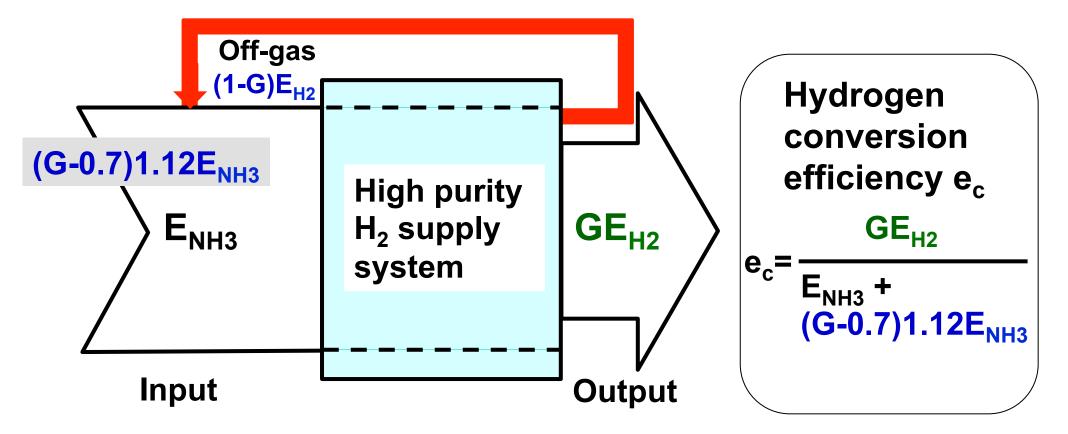
Purification efficiency: 70%(two-tower-type) $\rightarrow$  90%(four-tower-type) Hydrogen production rate: 1.3times (four-tower-type) Off-gas(10%): Heat source for ammonia decomposition Scale-up from 1Nm<sup>3</sup>/h to 10Nm<sup>3</sup>/h

# **Overview of 10Nm<sup>3</sup>/h-scale H<sub>2</sub> purifier (four-tower-type)**



We have a prospect for production of 300-1000Nm<sup>3</sup>/h H<sub>2</sub> purifier.

# Energy balance of ammonia decomposition and high purity H<sub>2</sub> supply system



E<sub>NH3</sub>: Heat of combustion for ammonia, 17.1MJ/(1Nm<sup>3</sup>/hNH<sub>3</sub>)

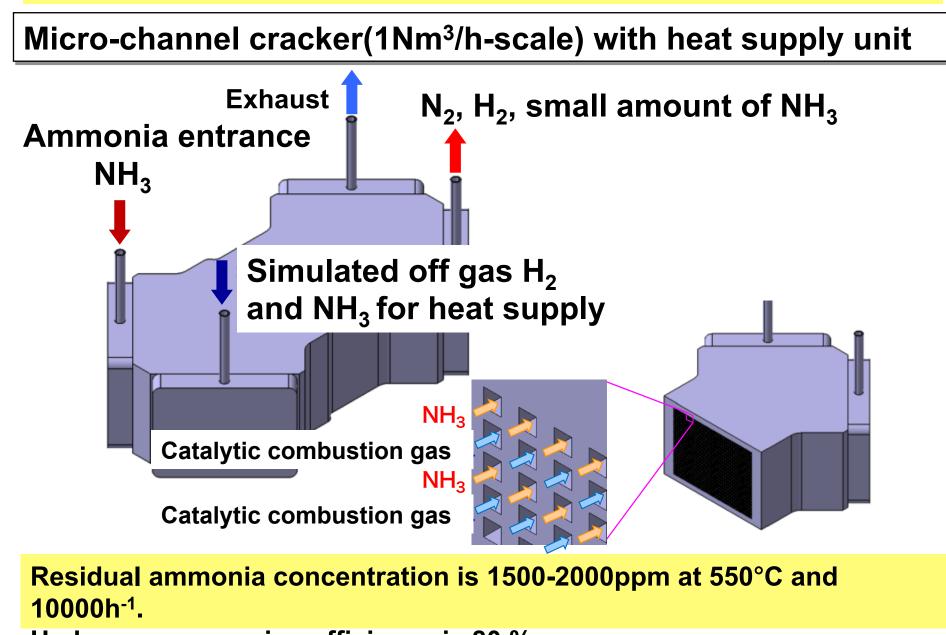
(G-0.7)1.12E<sub>NH3</sub>: Additional heat of decomposition

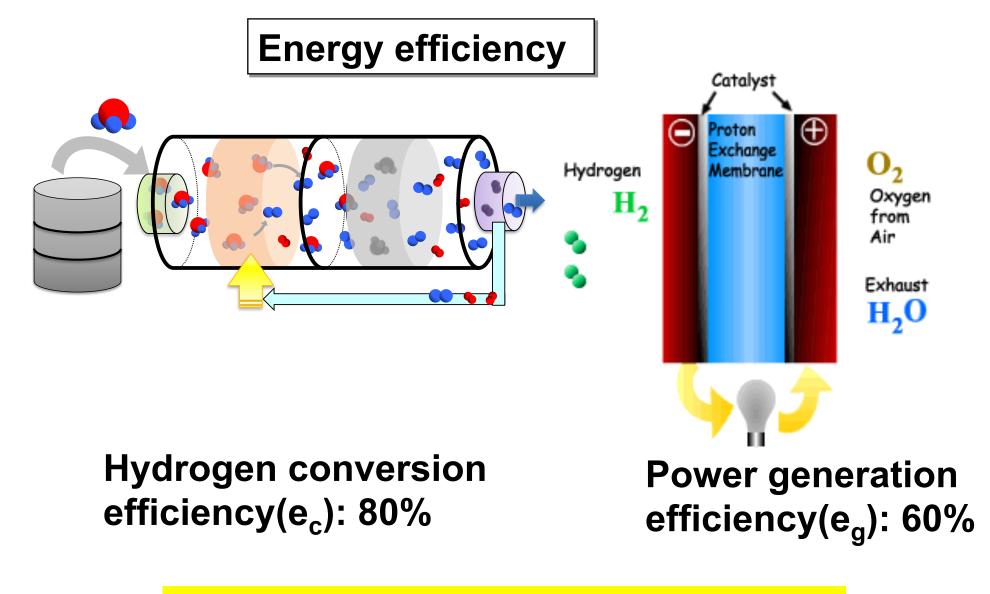
G: Hydrogen purification efficiency E<sub>H2</sub>: Heat of combustion for hydrogen 19.2MJ/(1.5Nm<sup>3</sup>/hH<sub>2</sub>)

Calculated hydrogen conversion efficiency: 82%(G=0.9) >78%(G=0.7)

#### (3) NH<sub>3</sub> cracker with heat supply unit

Micro-channel: High heat transfer performance compared with shell and tube





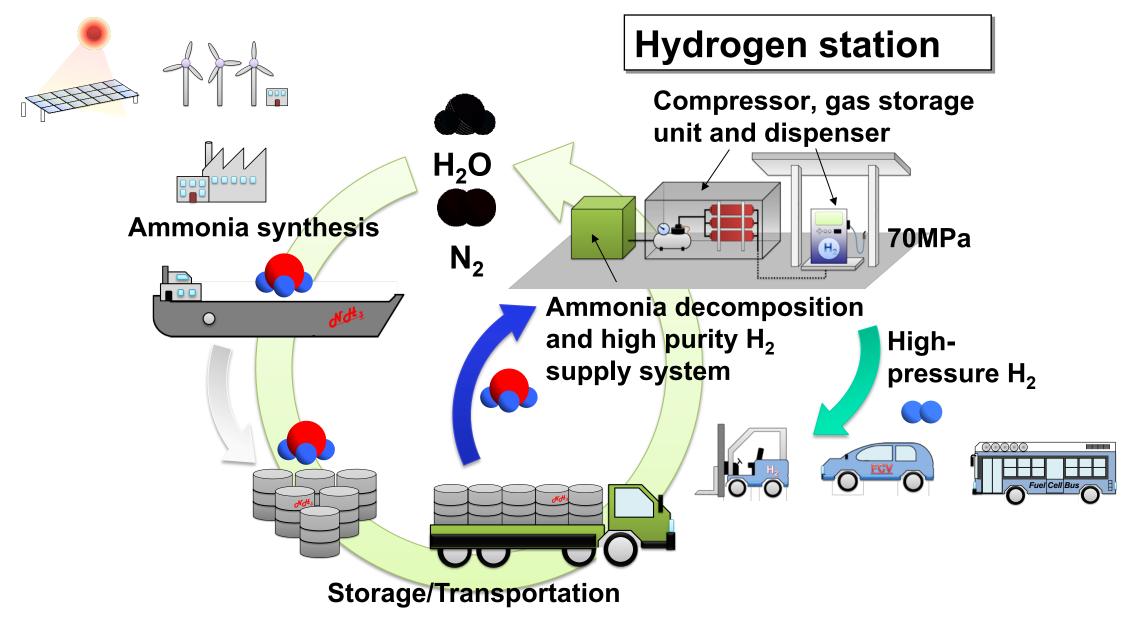
Energy efficiency(e<sub>c</sub>×e<sub>g</sub>): About 50%

# 3. Summary

- (1)We have developed component technologies to detect ammonia concentration below 0.1ppm.
- (2) We have developed 10Nm<sup>3</sup>/h-scale high efficiency H<sub>2</sub> purifier.
  (3) Hydrogen purification efficiency (hydrogen recovery rate) was 90% and production amount was 1.3 times of the conventional purifier.
- (4) Hydrogen conversion efficiency using micro-channel cracker was 80% and similar to the calculated value (82%).

#### Acknowledgements

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# Thank you for your attention.